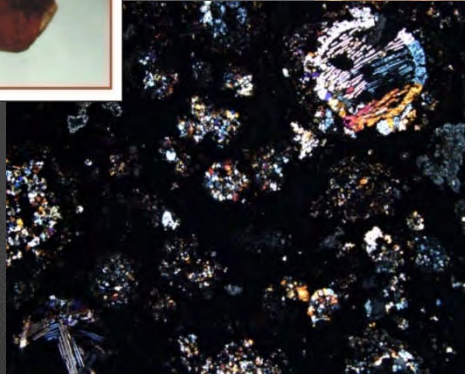


# Mineralogical Co-Evolution of the Geo- and Biospheres



Mid-Atlantic Senior Physicists Group  
October 17, 2012

Robert M. Hazen, Geophysical Laboratory





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## **Colorado State**

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Aaron Zimmerman

## **Univ. of Tennessee**

Linda Kah

# **Mineral Evolution: Outline**

- 1. What is mineral evolution?**
- 2. Ten stages of mineral evolution.**
- 3. Implications of mineral evolution.**
- 4. Recent discoveries in mineral evolution.**



# What Is Mineral Evolution?

A change over time in:

- The diversity of mineral species
- The relative abundances of minerals
- The compositional ranges of minerals
- The grain sizes and shapes of minerals

Hazen et al. (2008) *Amer. Mineral.* 93, 1693; Hazen et al. (2009) *Amer. Mineral.* 94, 1293;  
Hazen et al. (2010) *Elements* 6, #1, 9-46; Hazen et al. (2011) *Amer. Mineral.* 96, 953.



# What Is Mineral Evolution?

Focus exclusively on near-surface (<3 km depth) phases.

- Accessible to study on Earth
- Most likely to be observed on other planets and moons
- Direct interaction with biology

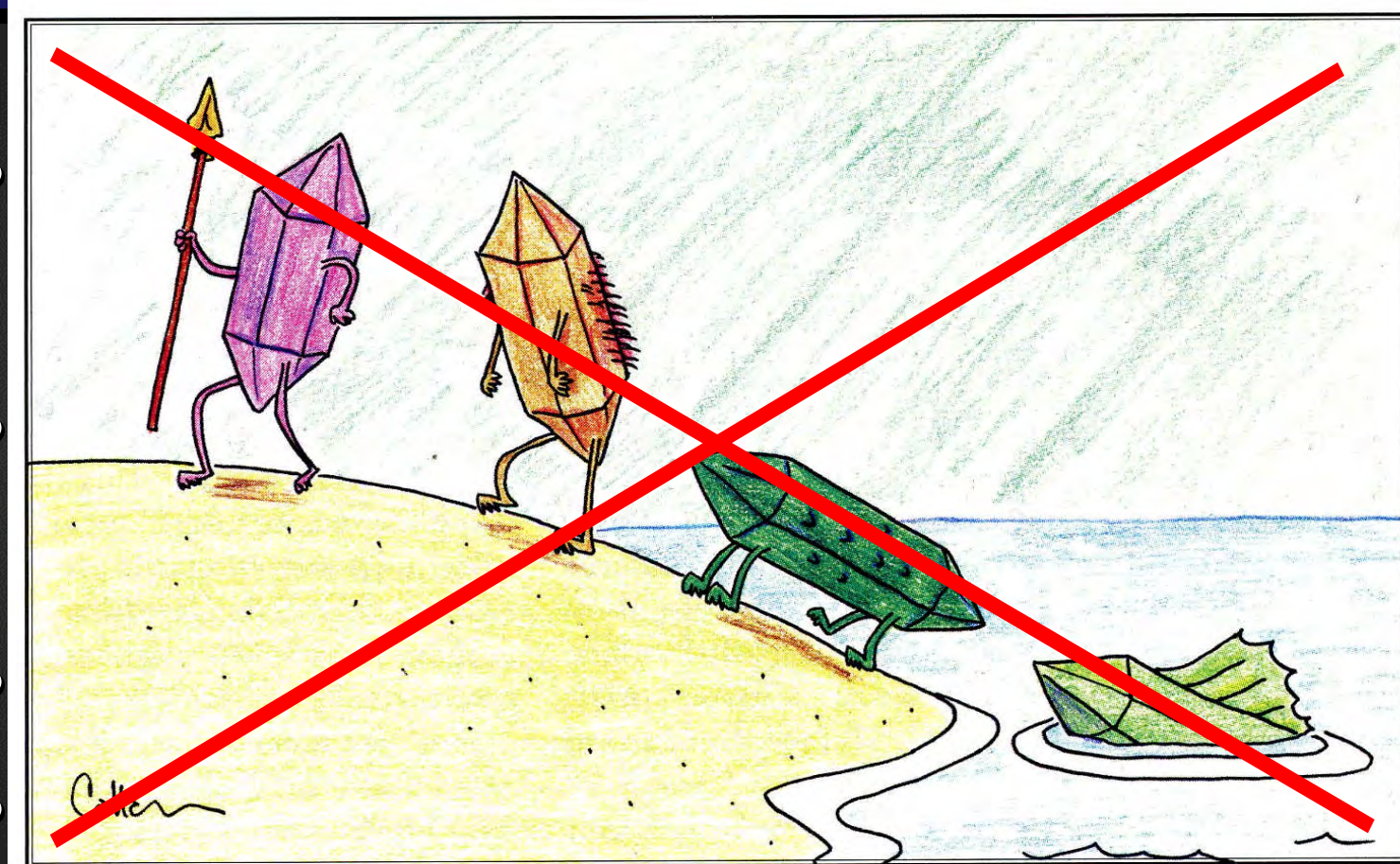
Hazen et al. (2008) *Amer. Mineral.* 93, 1693; Hazen et al. (2009) *Amer. Mineral.* 94, 1293;  
Hazen et al. (2010) *Elements* 6, #1, 9-46; Hazen et al. (2011) *Amer. Mineral.* 96, 953.

# **Why Mineral Evolution?**

- **Reframe mineralogy in a dynamic historical context**
- **Classify terrestrial planets and moons & identify mineralogical targets**
- **Explore general principles related to complex evolving systems**
- **Pose new mineralogical questions**



# A Comment on “Evolution”



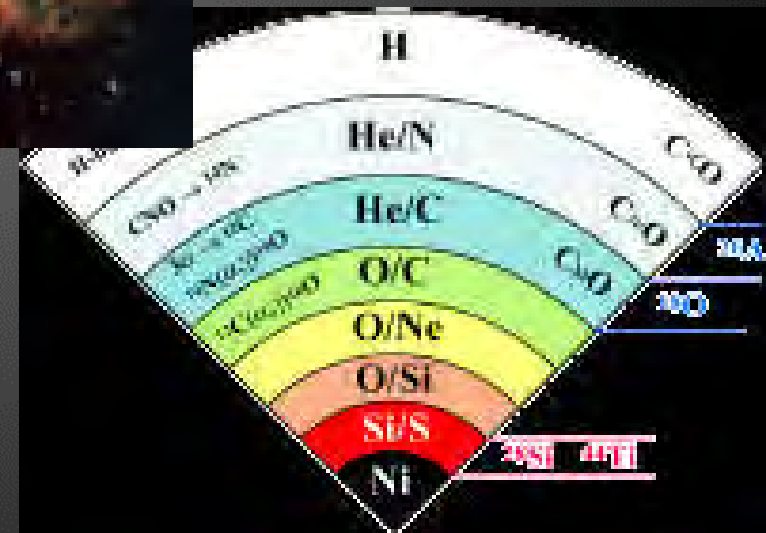
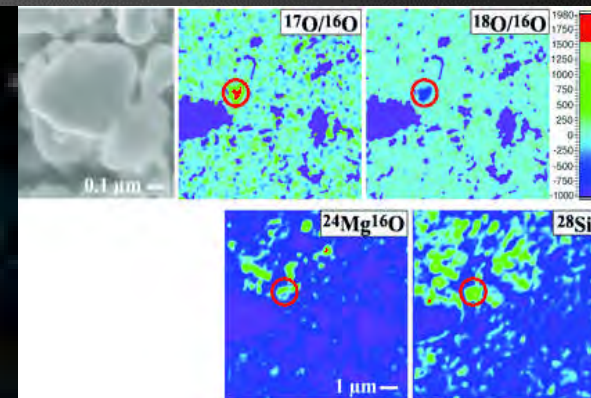
- But NOT Darwinian evolution!



# “Ur”-Mineralogy

Pre-solar grains contain about a dozen micro- and nano-mineral phases:

- Diamond/Lonsdaleite
- Graphite (C)
- Moissanite (SiC)
- Osbornite (TiN)
- Nierite (Si<sub>3</sub>N<sub>4</sub>)
- Rutile (TiO<sub>2</sub>)
- Corundum (Al<sub>2</sub>O<sub>3</sub>)
- Spinel (MgAl<sub>2</sub>O<sub>4</sub>)
- Hibbonite (CaAl<sub>12</sub>O<sub>19</sub>)
- Forsterite (Mg<sub>2</sub>SiO<sub>4</sub>)
- Nano-particles of TiC, ZrC, MoC, FeC, Fe-Ni metal within graphite.
- GEMS (silicate glass with embedded metal and sulfide).





# **Mineral Evolution:**

**How did we get from a dozen minerals to ~4500 on Earth today?**

**What does the distribution of minerals through time tell us about key tectonic, geochemical, and biological events?**

# **What Drives Mineral Evolution?**

**Deterministic and stochastic processes  
that occur on any terrestrial body:**

- 1. The progressive separation and concentration of chemical elements from their original uniform distribution.**



# What Drives Mineral Evolution?

Deterministic and stochastic processes that occur on any terrestrial body:

1. The progressive separation and concentration of chemical elements from their original uniform distribution.
2. **An increase in the range of intensive variables (T, P, activities of volatiles).**

# What Drives Mineral Evolution?

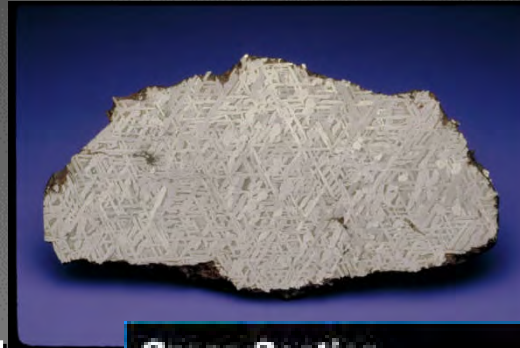
Deterministic and stochastic processes that occur on any terrestrial body:

1. The progressive separation and concentration of chemical elements from their original uniform distribution.
2. An increase in the range of intensive variables (T, P, activities of volatiles).
3. **The generation of far-from-equilibrium conditions by living systems.**

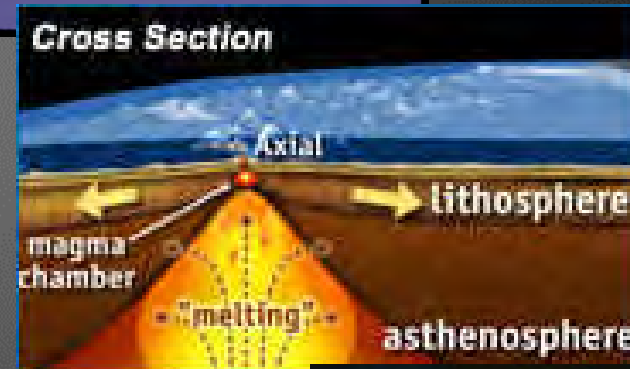


# Three Eras of Earth's Mineral Evolution

1. The Era of Planetary Accretion



2. The Era of Crust and Mantle Reworking



3. The Era of Bio-Mediated Mineralogy

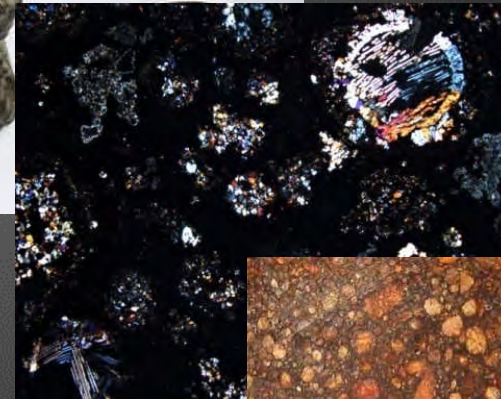


# Stage 1: Primary Chondrite Minerals

Minerals formed ~4.56 Ga in the Solar nebula  
“as a consequence of condensation, melt  
solidification or solid-state recrystallization”  
(MacPherson 2007)

~60 mineral species

- CAIs
- Chondrules
- Silicate matrix
- Opaque phases

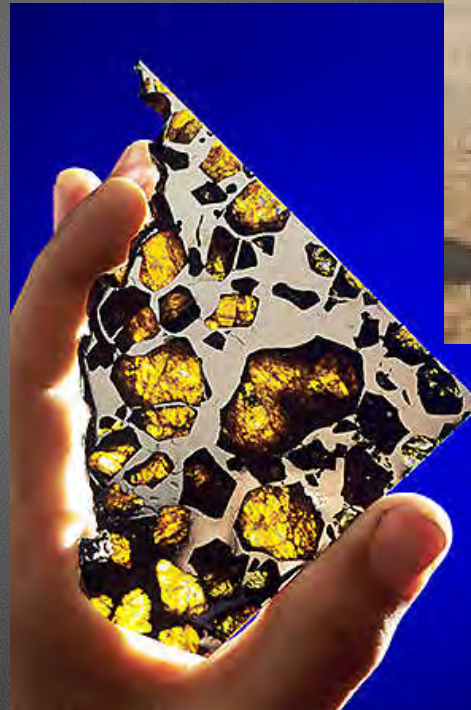




## Stage 2: Aqueous alteration, metamorphism and differentiation of planetesimals

**~250 mineral known species: 4.56-4.55 Ga**

- First albite & K-spar
- First significant  $\text{SiO}_2$
- Feldspathoids
- Hydrous biopyriboles
- Clay minerals
- Zircon
- Shock phases



## Stage 2: Planetary Accretion





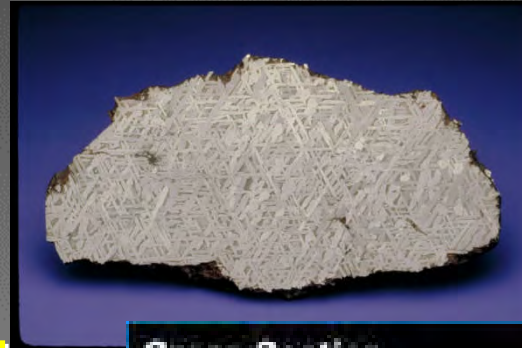
## **Stages 1 and 2: Planetary Accretion**

**In these early stages all of Earth's near-surface compositional complexity was present, but it was not manifest in a diversity of unusual mineral species.**

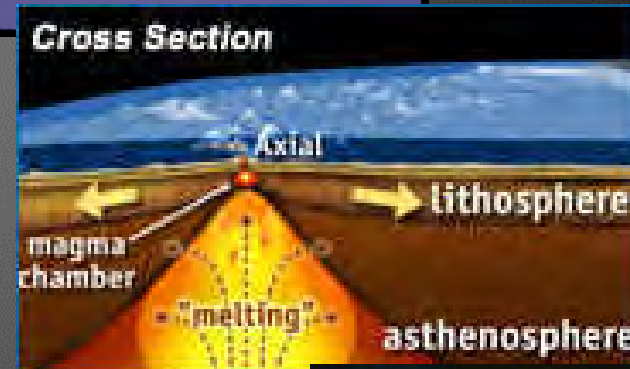
**>250 mineral species**

# Three Eras of Earth's Mineral Evolution

1. The Era of Planetary Accretion



2. The Era of Crust and Mantle Reworking

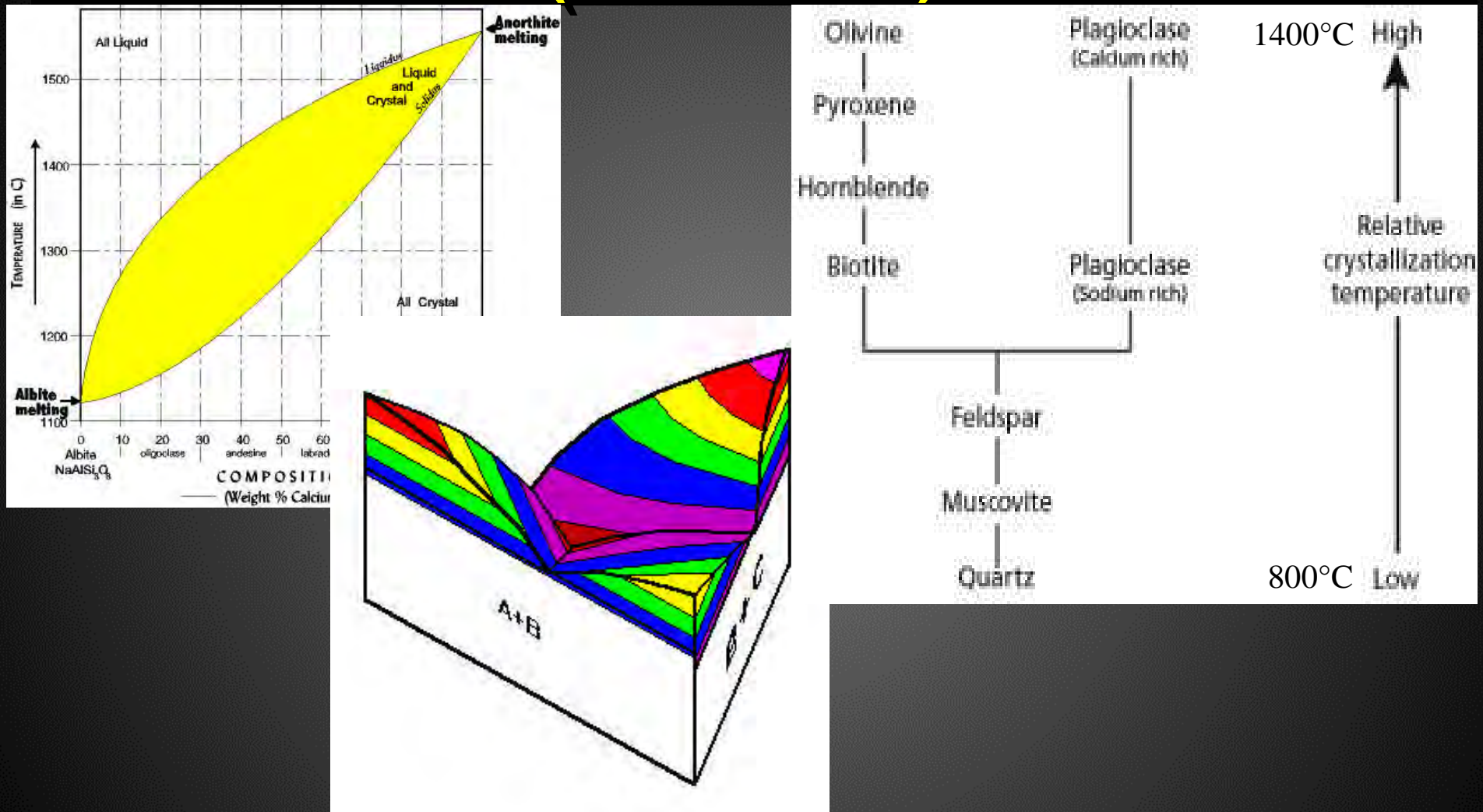


3. The Era of Bio-Mediated Mineralogy





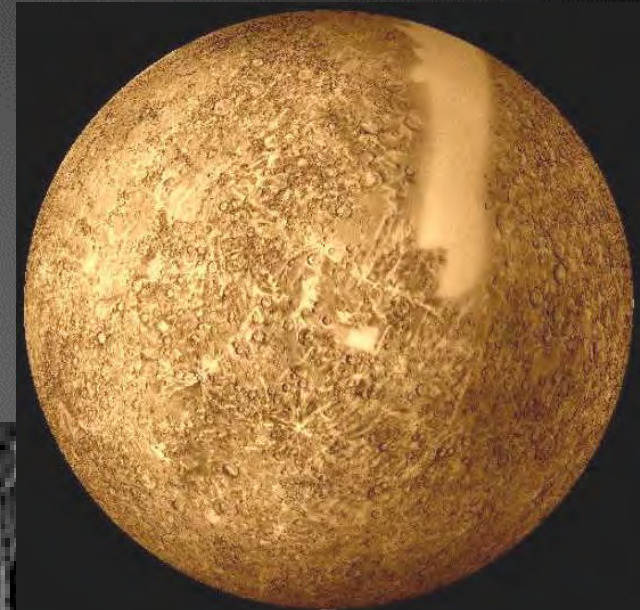
# Stage 3: Initiation of Igneous Rock Evolution (4.55-4.0 Ga)



**Partial melting, fractional crystallization  
and magma immiscibility**

# Stage 3: Initiation of Igneous Rock Evolution Volatile-poor Body

**~350 mineral species?**



**Is this the end point of the Moon and Mercury?**



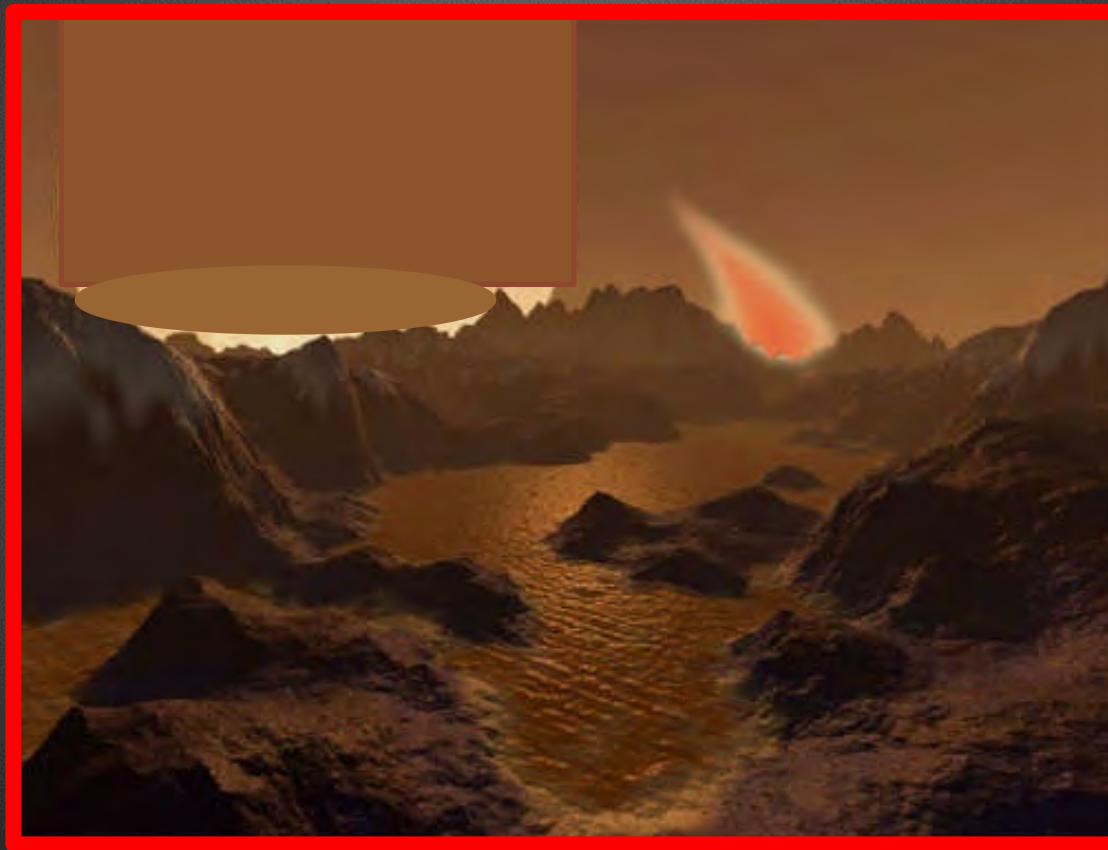
## Stage 3: Initiation of Igneous Rock Evolution on a Volatile-rich Body (4.55-4.0 Ga)



**Volcanism, outgasing and surface hydration.**

## Stage 3: Initiation of Igneous Rock Evolution Volatile-rich Body

**>500 mineral species (hydroxides, clays)**



**Volcanism, outgasing, surface hydration, evaporites, ices.**



# The Formation of the Moon



## Stage 3: Initiation of Igneous Rock Evolution Volatile-rich Body

**>500 mineral species (hydroxides, clays)**

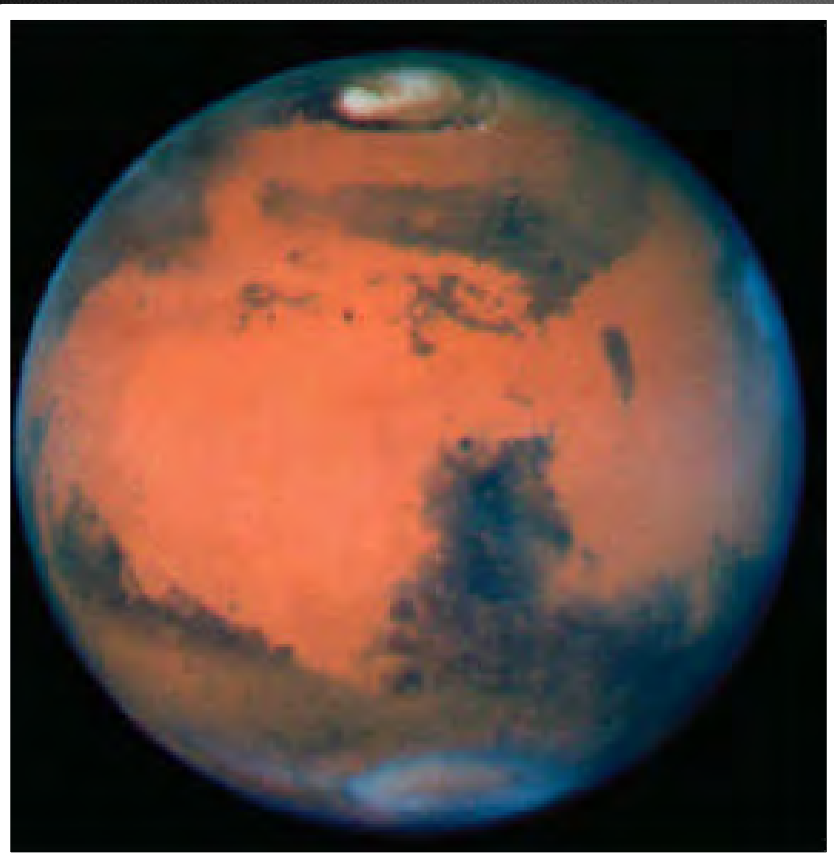


**Volcanism, outgasing, surface hydration, evaporites, ices.**



## Stage 3: Initiation of Igneous Rock Evolution Volatile-rich Body

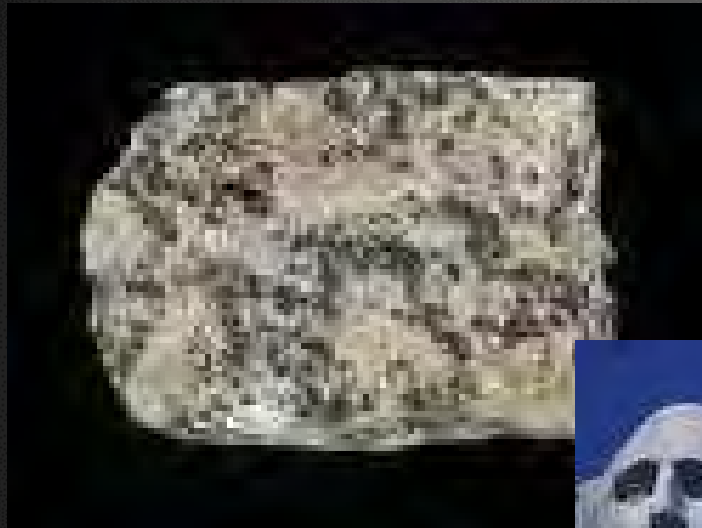
**Is this the end point for Mars?**



**Volcanism, outgasing, surface hydration, evaporites, ices.**

## Stage 4: Granitoid Formation (>3.5 Ga)

>1000 mineral species (pegmatites)

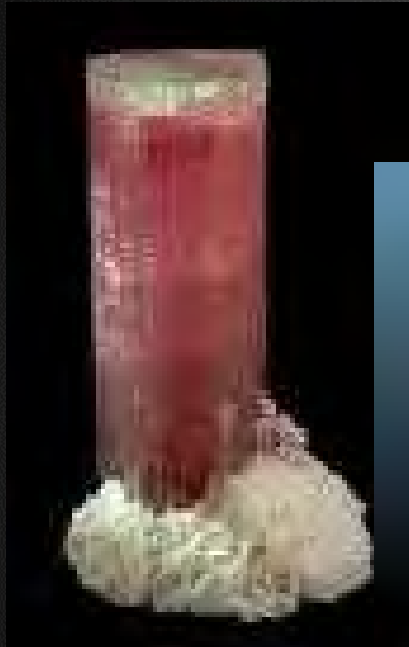


Partial melting of basalt and/or sediments.

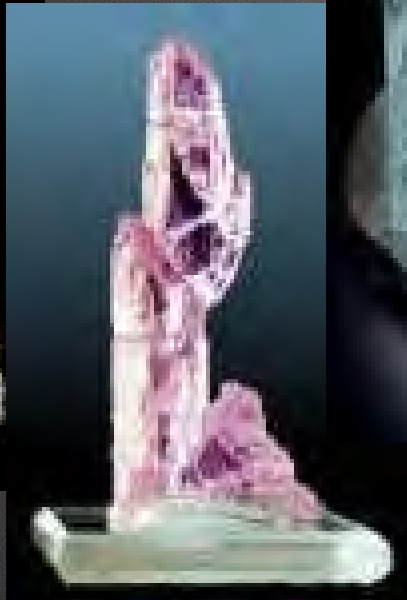


## Stage 4: Granitoid Formation (>3.5 Ga)

>1000 mineral species (pegmatites)



Tourmaline



Spodumene



Beryl



Pollucite

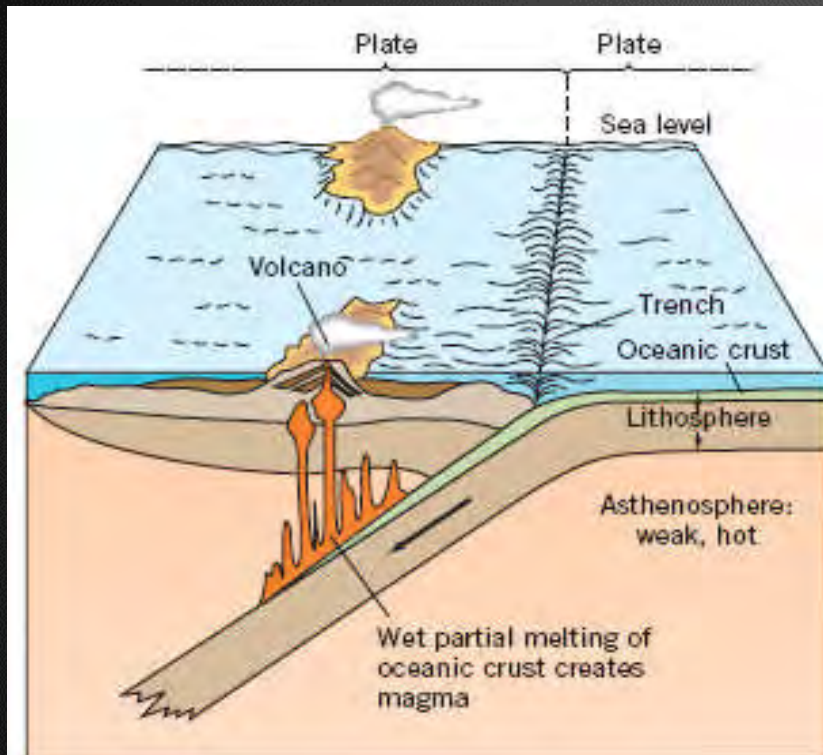


Tantalite

Complex pegmatites require multiple cycles of eutectic melting and fluid concentration.

Must they be younger than 3.5 Ga?

## Stage 5: Plate tectonics and large-scale hydrothermal reworking of the crust (>3 Ga)



$\sim 10^8 \text{ km}^3$  of reworking

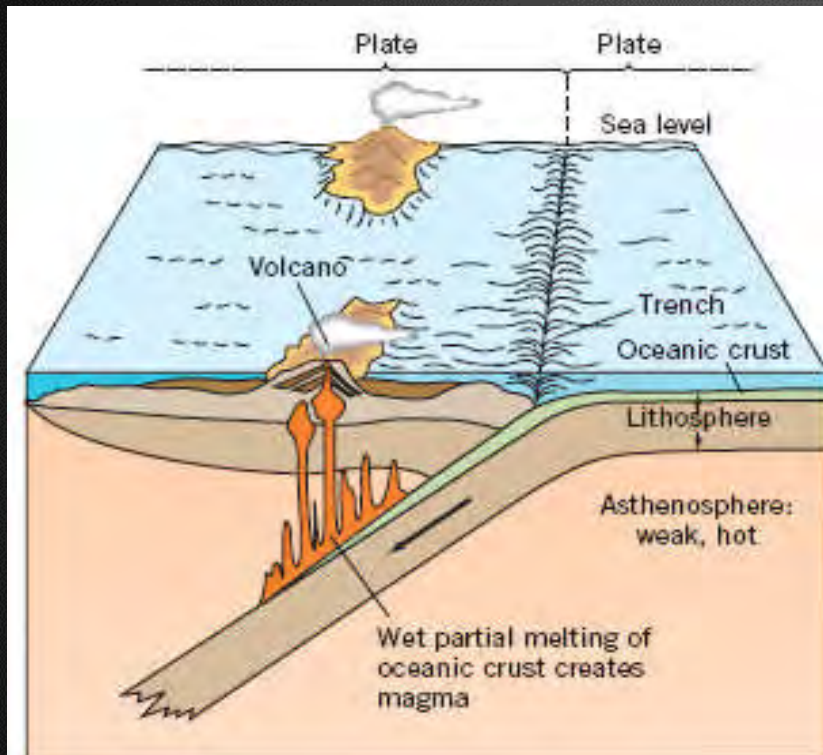


Mayon Volcano, Philippines

## New modes of volcanism



## Stage 5: Plate tectonics and large-scale hydrothermal reworking of the crust (>3 Ga)



Rio Tinto. Spain

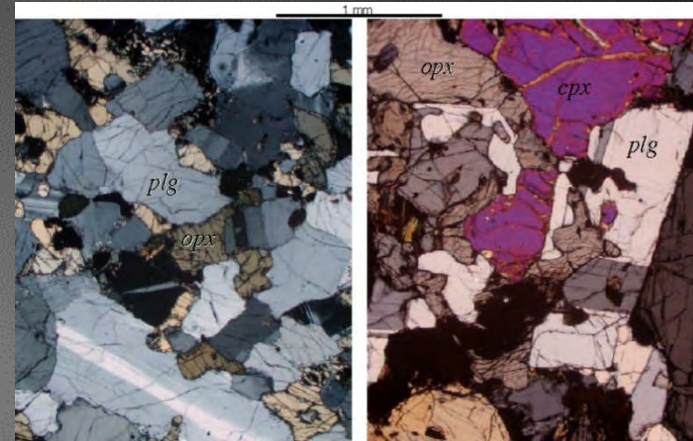
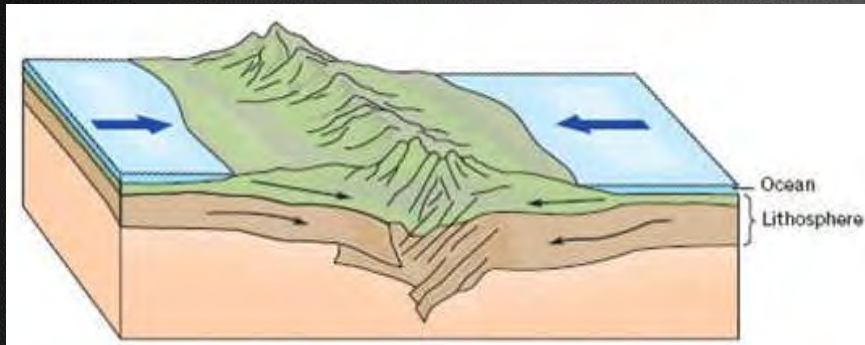
**New modes of volcanism**

**Massive base metal deposits (sulfides, sulfosalts)**

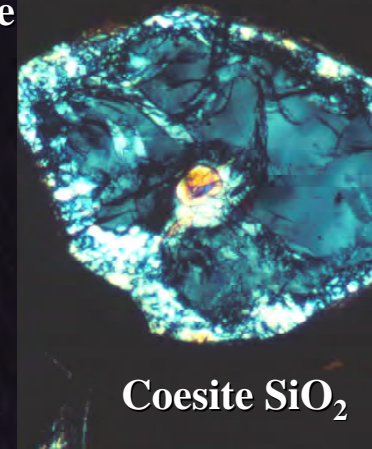


# Stage 5: Plate tectonics and large-scale hydrothermal reworking of the crust (>3 Ga)

1,500 mineral species



Glaucophane, Lawsonite, Jadeite

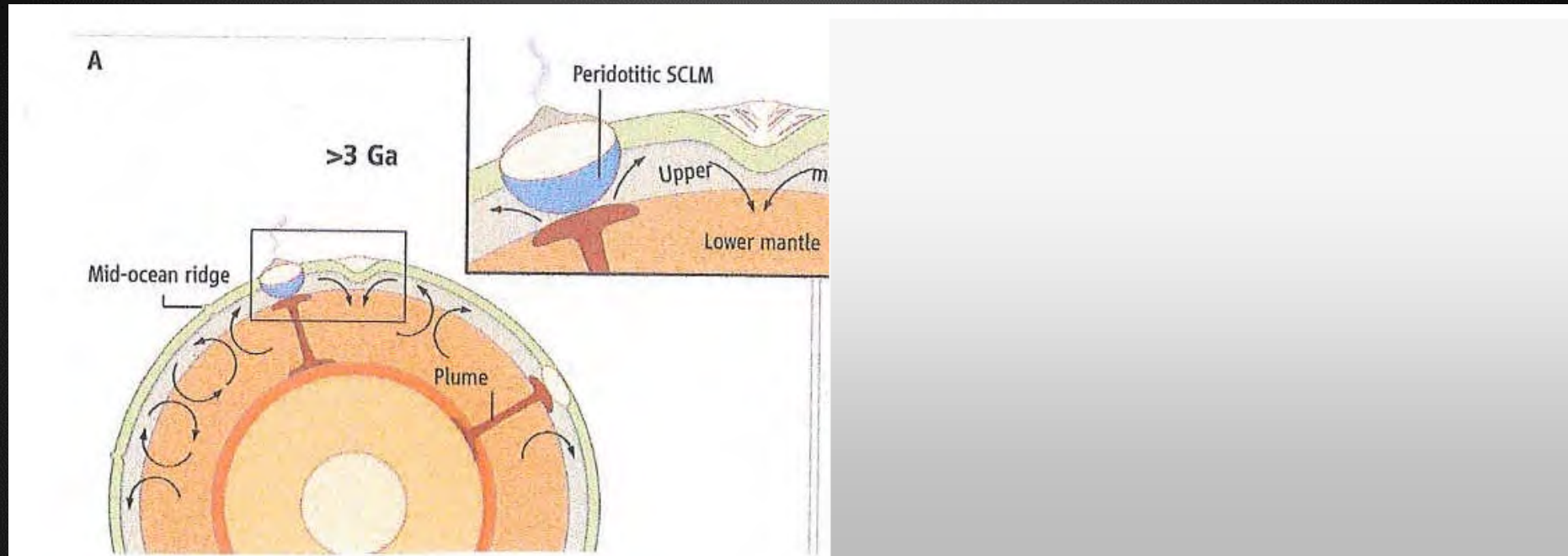


Coesite  $\text{SiO}_2$

High-pressure metamorphic suites  
(blueschists; granulites; UHP phases)



# Stage 5: Plate tectonics and large-scale hydrothermal reworking of the crust (>3 Ga)

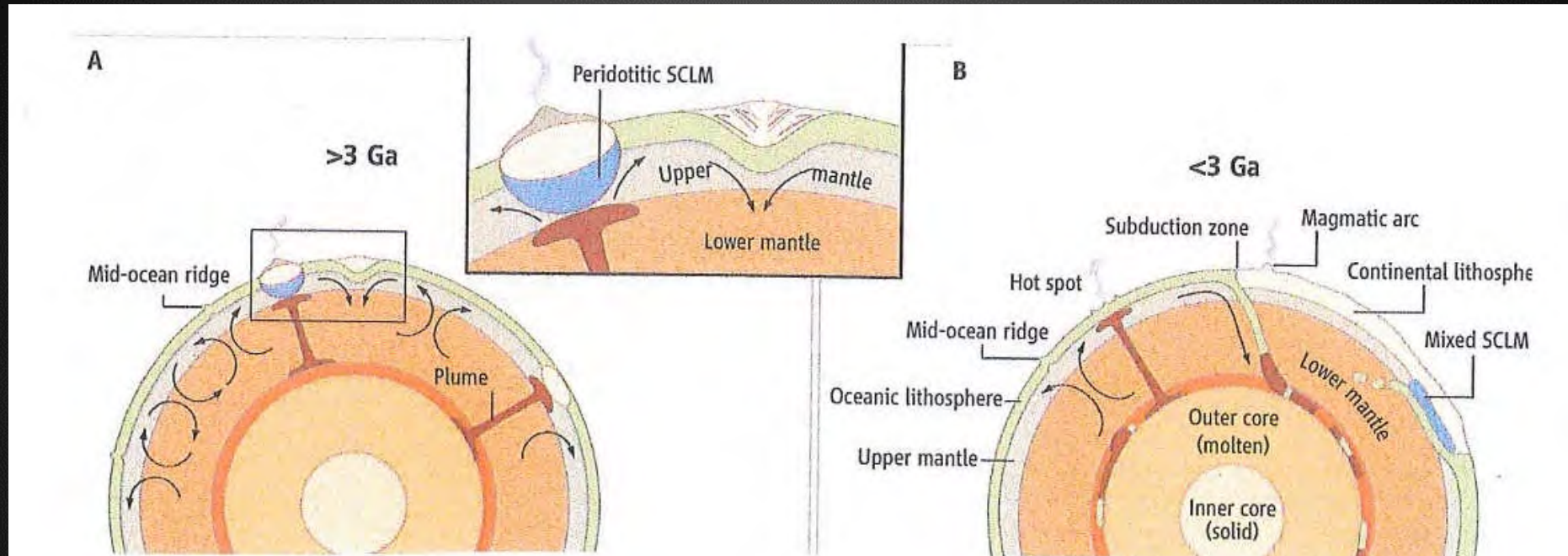


Van Kranendonk (2011)

> 3 Ga

When did subduction begin?

# Stage 5: Plate tectonics and large-scale hydrothermal reworking of the crust (>3 Ga)



Van Kranendonk (2011)

> 3 Ga

< 3 Ga

When did subduction begin?



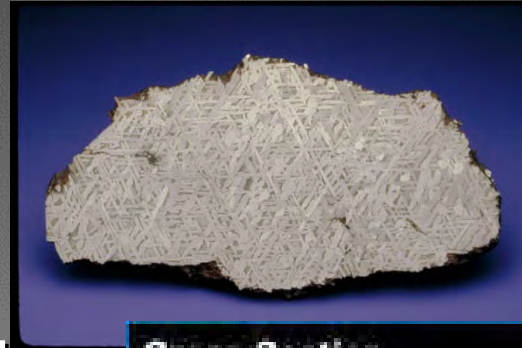
## **Stages 3-5: Era of crust-mantle processing (igneous evolution; plate tectonics)**

**New geologic processes, especially fluid-rock interactions associated with igneous activity and plate tectonics, led to a greater diversity of geochemical environments and thus new mineral species.**

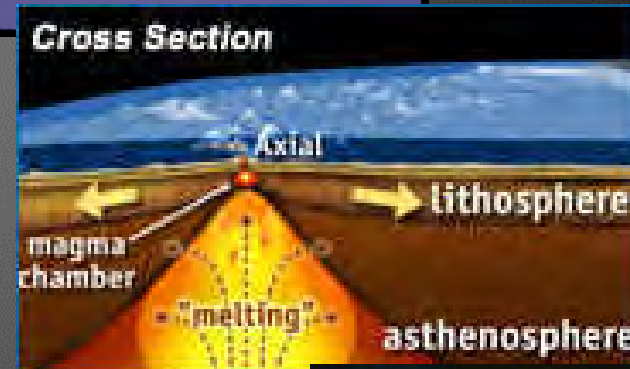
**~1500 mineral species**

# Three Eras of Earth's Mineral Evolution

## 1. The Era of Planetary Accretion



## 2. The Era of Crust and Mantle Reworking



## 3. The Era of Bio-Mediated Mineralogy





**The origin of life may require some minimal degree of mineral evolution.**

**Sulfides**



**Clays**



**Borates**



**Conversely, does further mineral evolution depend on life?**

**Hence the co-evolution of the geo- and biospheres.**



## Stage 6: Anoxic Archean biosphere (3.9-2.5 Ga)

**~1,500 mineral species** (BIFs, carbonates,



**Photo credit: D. Papineau**



**Photo credit: F. Corsetti, USC**



## Stage 6: Anoxic Archean biosphere (3.9-2.5 Ga)

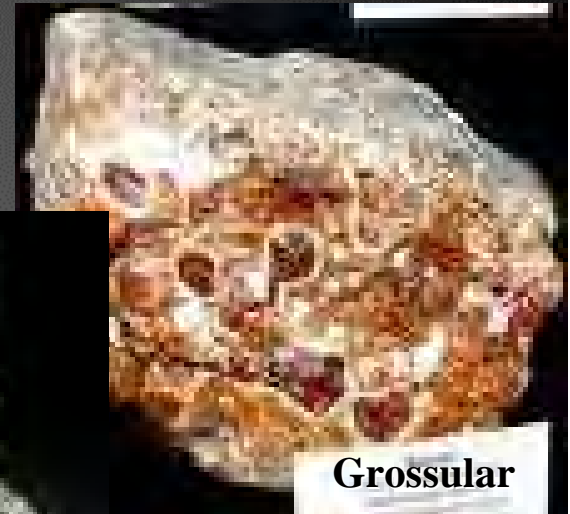
**~1,500 mineral species** (BIFs, carbonates, sulfates, evaporites, skarns)



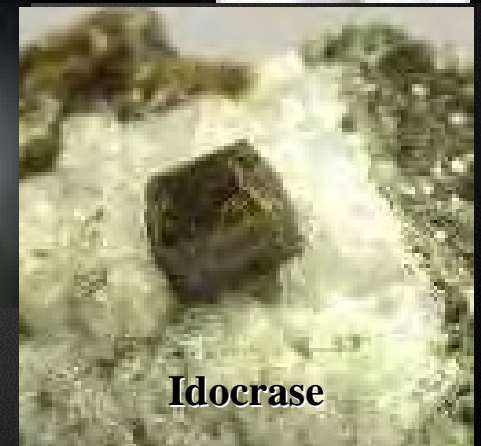
Death Valley evaporites  
(courtesy Smith College)



Diopside



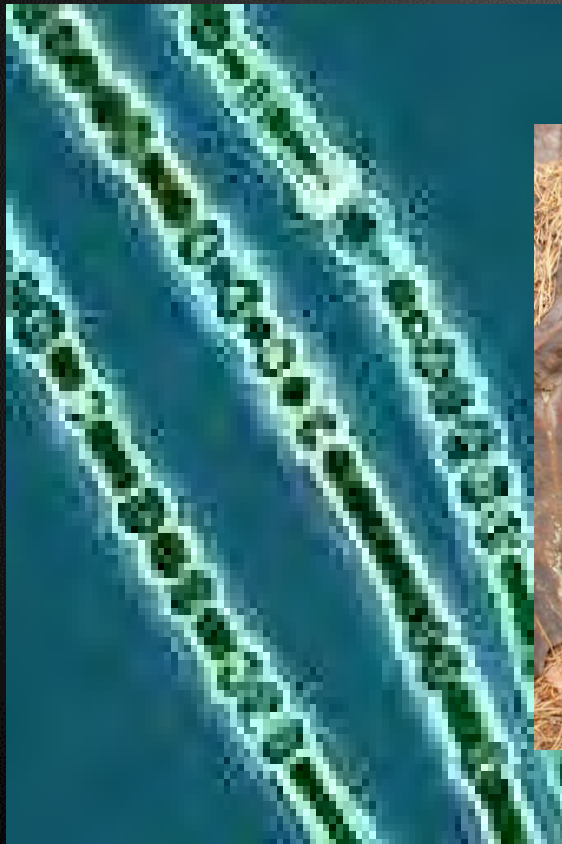
Grossular



Idocrase

## Stage 7: Paleoproterozoic Oxidation (2.5-1.85 Ga)

>4000 mineral species, including perhaps  
>2,000 new oxides/hydroxides



Rise of oxidative photosynthesis.



## Stage 7: Paleoproterozoic Oxidation (2.5-1.85 Ga)

**>4000 mineral species, including perhaps  
>2,000 new oxides/hydroxides**



**Rise of oxidative photosynthesis.**

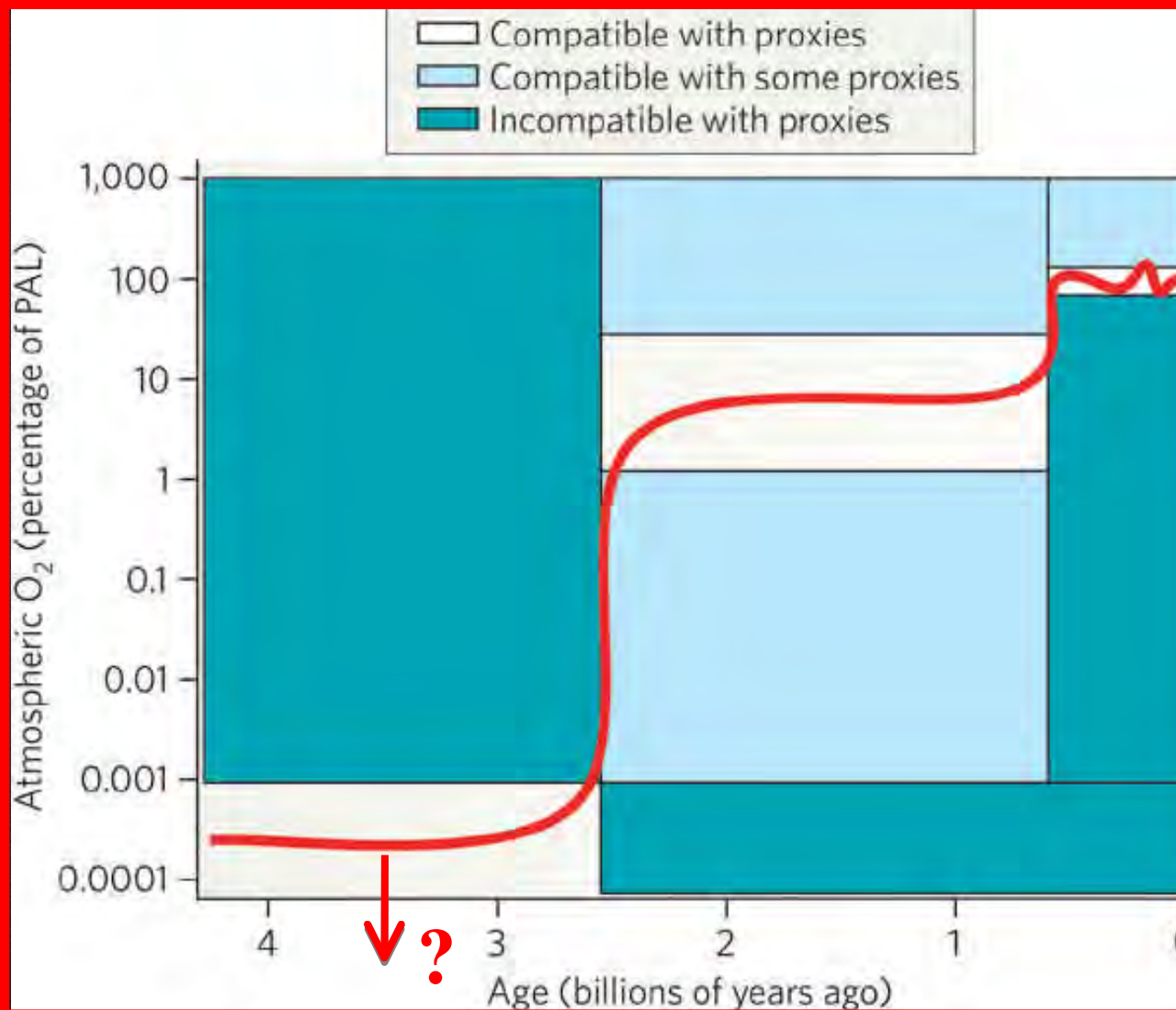
# Hypothesis

**Approximately 2/3rds of all known mineral species cannot form in an anoxic environment, and thus are the indirect consequence of biological activity.**

**Many lines of evidence point to an essentially anoxic Archean atmosphere.**



# What was the oxygen fugacity in the Archean?



Kump (2008) *Nature* 451, 277-278.

# What was the oxygen fugacity in the Archean?

## Published estimates of Archean log fO<sub>2</sub>

Ohmoto (numerous refs)	> -2
Farquhar et al. (2000)	< -5
Frimmel (2005)	< -5
Kump (2008)	< -5
C-W-K-H Model (1968+)	~ -13
Sverjensky et al. (2008, 2010)	~ -70



# **Key constraints on Archean surface oxygen fugacity.**

**Detrital uraninite and pyrite**

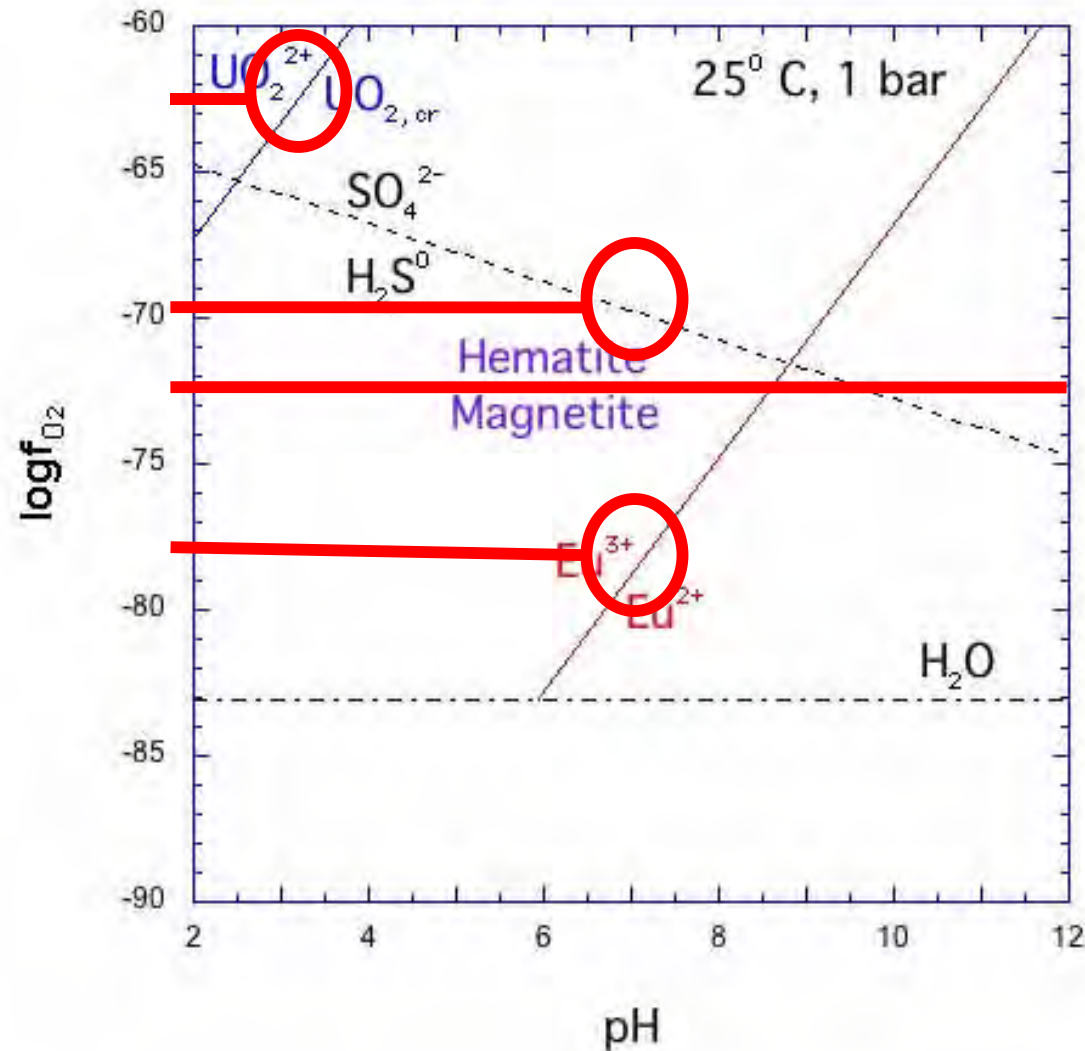
**Paleosols lacking iron oxides**

**[Surface waters with aqueous  $\text{Fe}^{2+}$ ]**

**[Surface waters with low  $\text{SO}_4^{2-}$ ]**

**$\text{Eu}^{2+}$  anomalies**

# What was the oxygen fugacity in the Archean?





## **Key constraints on Archean surface oxygen fugacity.**

**Detrital uraninite, pyrite and siderite**

**Paleosols lacking iron oxides**

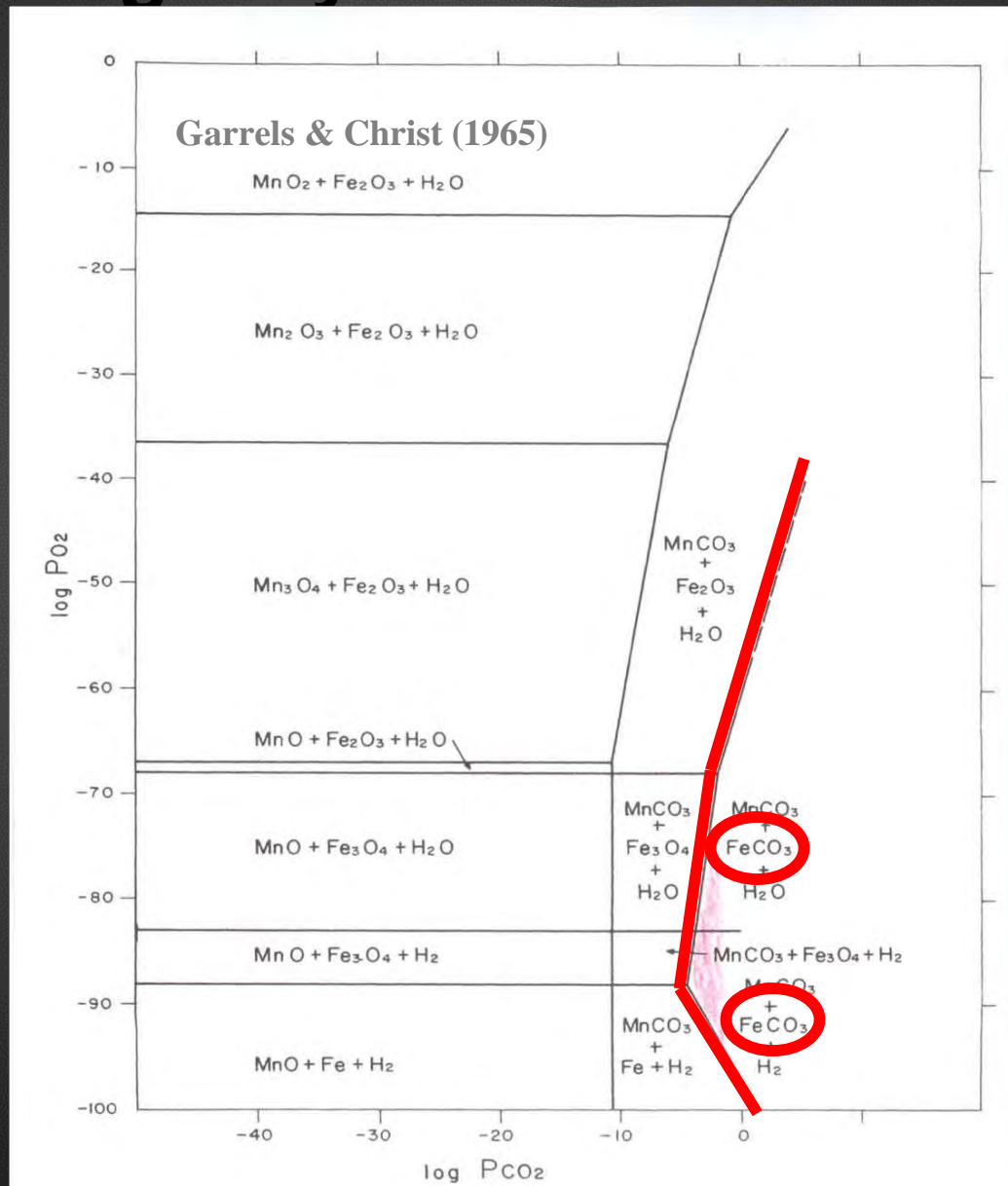
**[Surface waters with aqueous  $\text{Fe}^{2+}$ ]**

**[Surface waters with low  $\text{SO}_4^{2-}$ ]**

**$\text{Eu}^{2+}$  anomalies**

**Precipitation of ferroan carbonates**

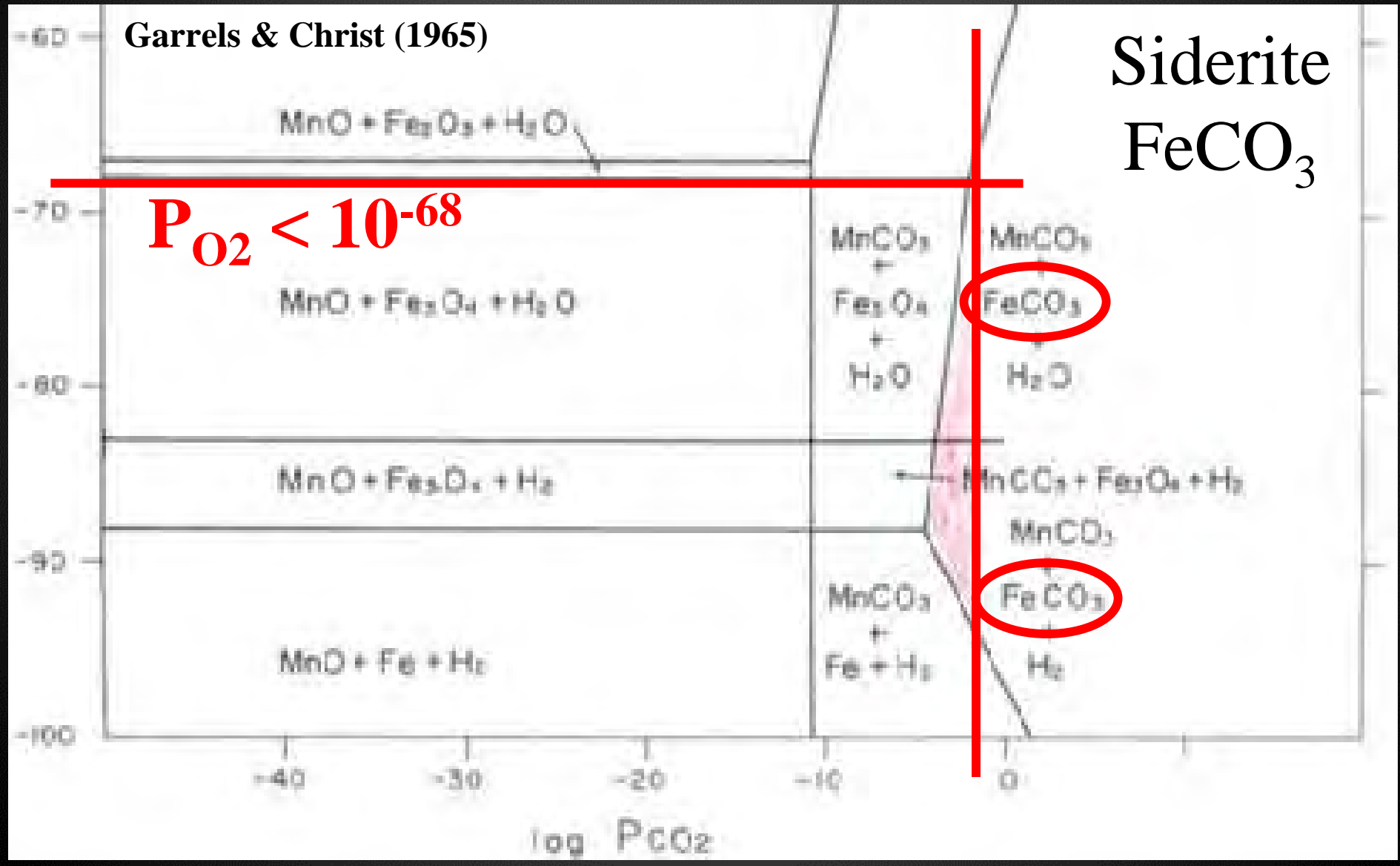
# What was the oxygen fugacity in the Archean?



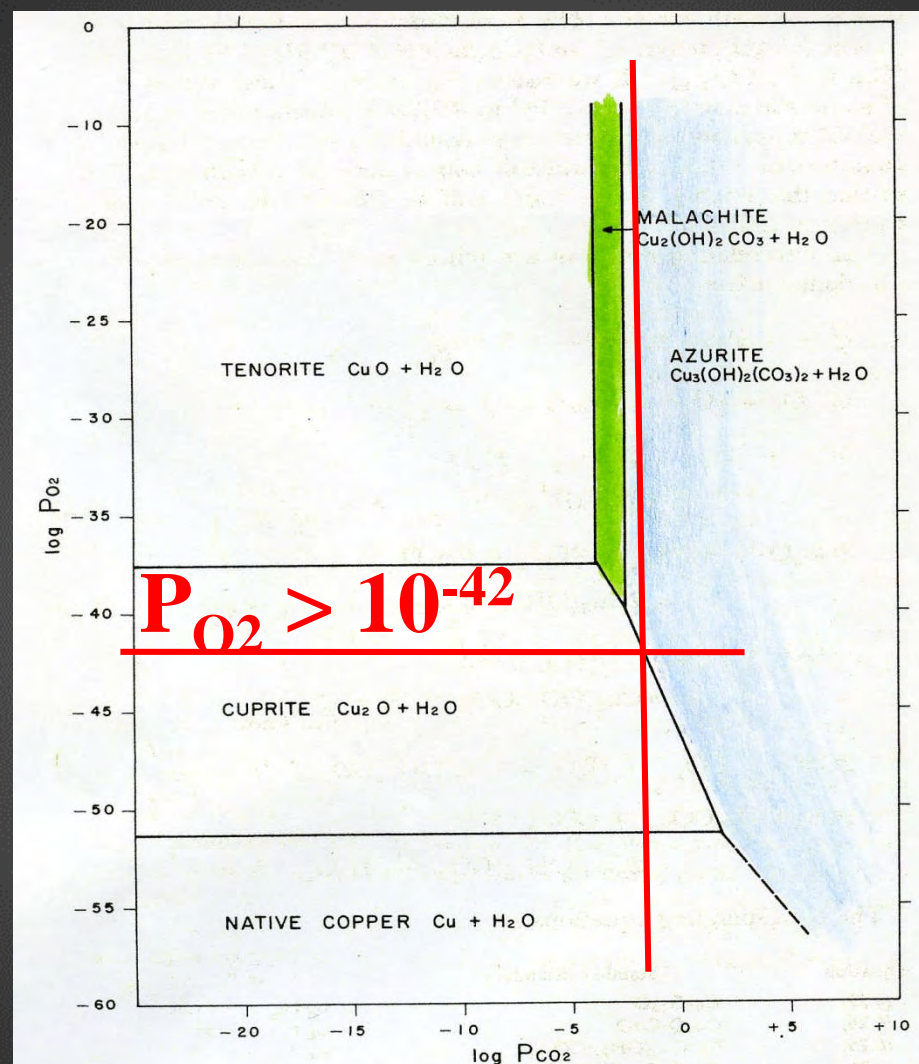
Siderite  
 $FeCO_3$



# What was the oxygen fugacity in the Archean?



# What minerals won't form?

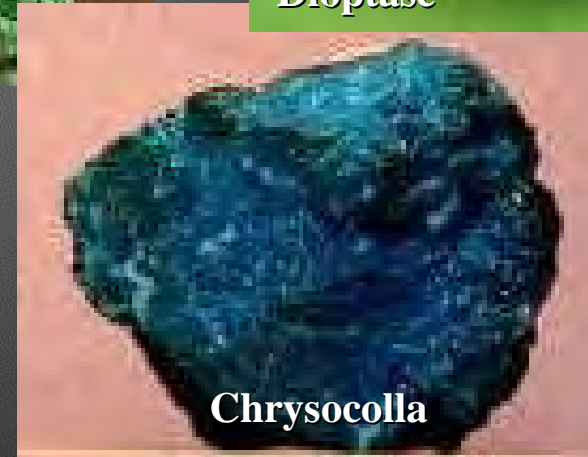
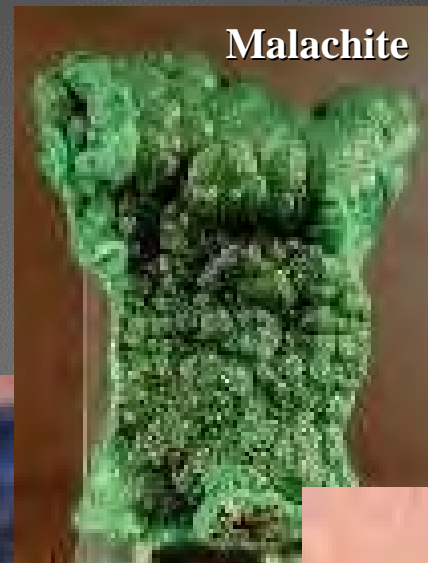


If the effective  $\log f_{O_2} \sim -70$ , then malachite, azurite and other  $Cu^{2+}$  minerals will not form.



## Stage 7: Paleoproterozoic Oxidation (2.5-1.85 Ga)

$\text{Cu}^{2+}$  Copper minerals (256 of 321)



When did these minerals first appear?

## Stage 7: Paleoproterozoic Oxidation (2.5-1.85 Ga)

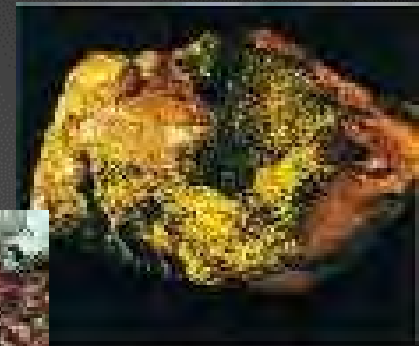
What mineral species won't form?

02 of 220 U minerals

9 of 451 Mn minerals

7 of 56 Ni minerals

32 of 790 Fe minerals



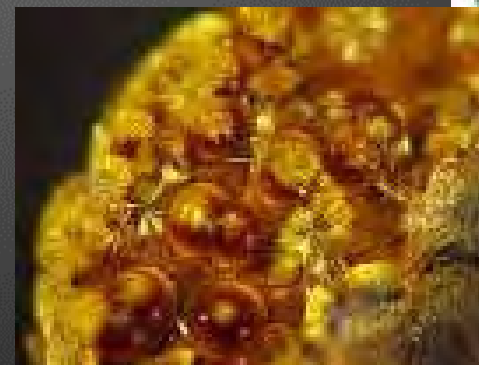
CARNOTITE



Piemontite



Garnierite



Xanthoxenite



## **Stages 6-10: Co-evolution of the geosphere and biosphere**

**Changes in Earth's atmospheric composition at ~2.4 to 2.2 Ga represent the single most significant factor in our planet's mineralogical diversity.**

**>4500 mineral species**

# RESULTS: Molybdenite ( $\text{MoS}_2$ ) through Time

— Golden et al. (submitted)

SE HERE



GOE HERE



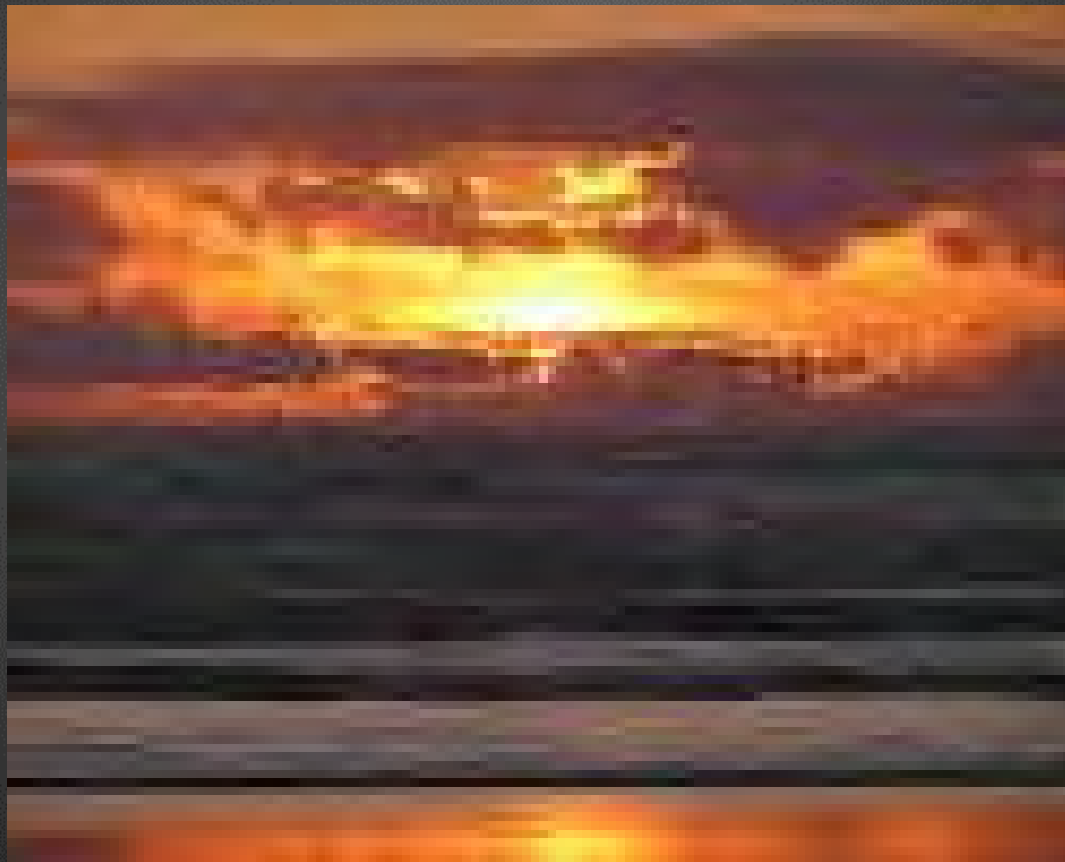


## RESULTS: Molybdenite ( $\text{MoS}_2$ ) through Time

**Hypothesis: There was a protracted “Great Subsurface Oxidation Interval” that postdated the GOE by a billion years. This interval was the single most significant factor in Earth’s mineralogical diversification.**

## Stage 8: The “Intermediate Ocean” (1.85-0.85 Ga)

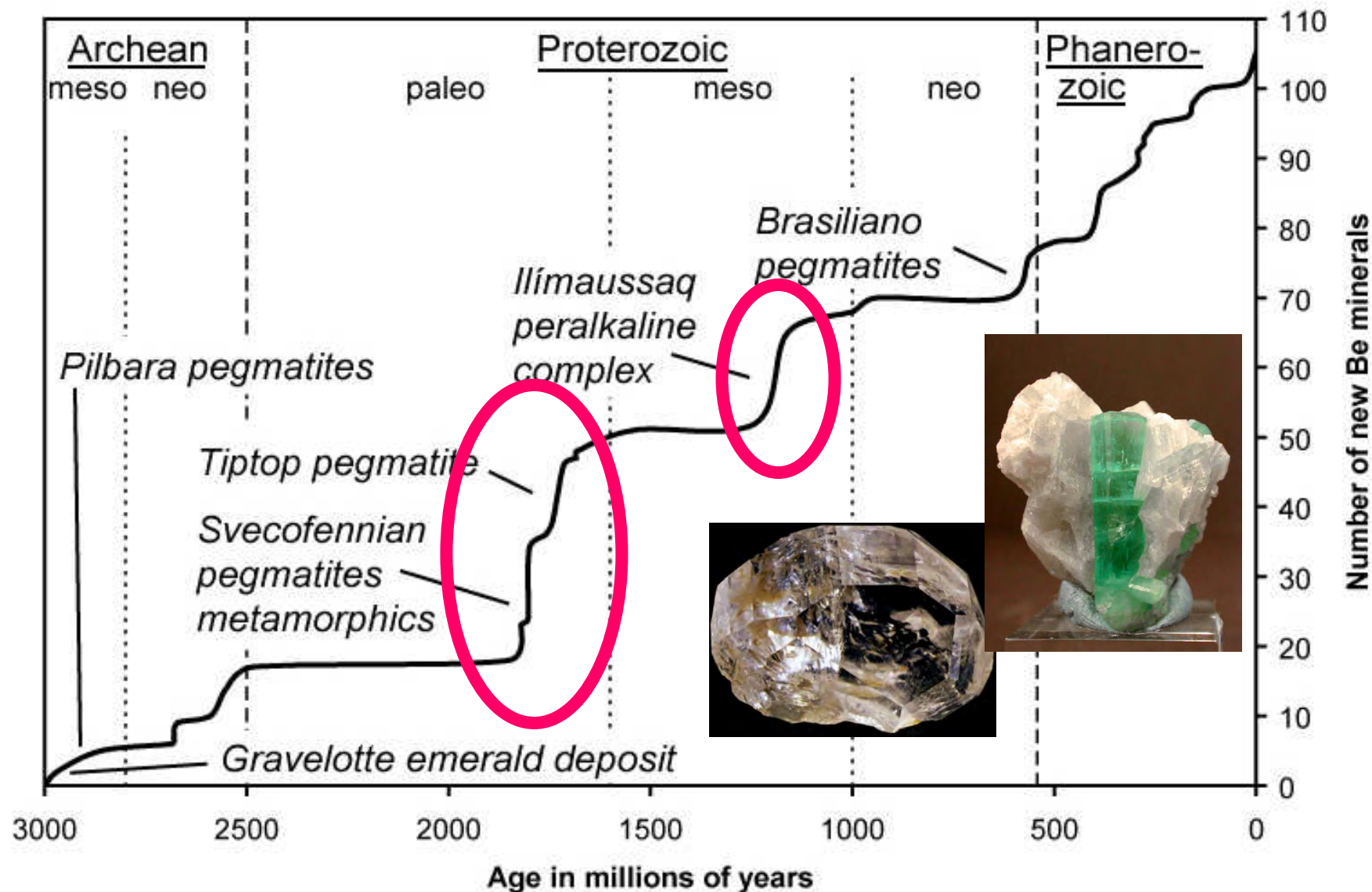
**>4000 mineral species (few new species)**



**Oxidized surface ocean; deep-ocean anoxia.**

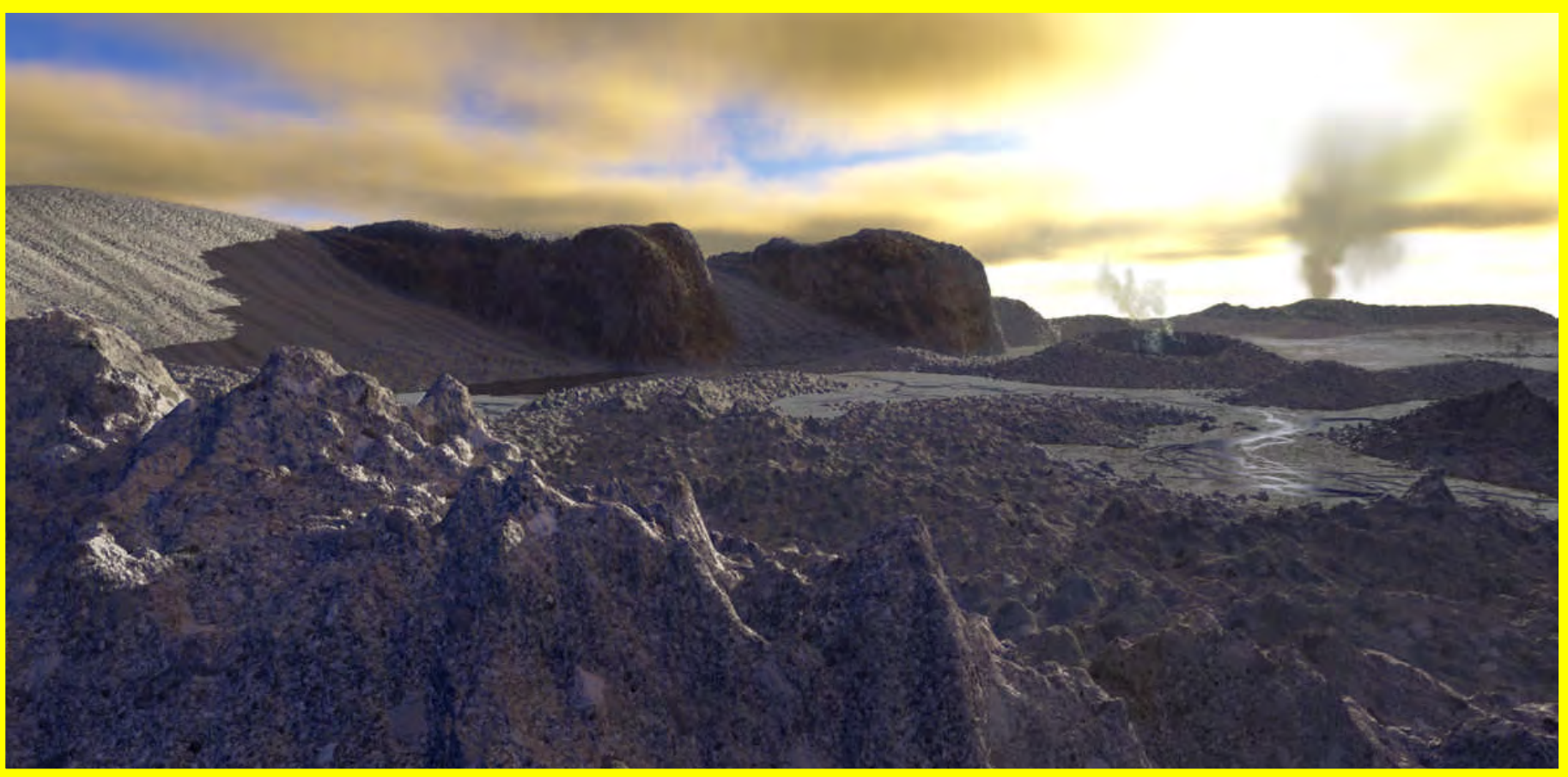


# Be Mineral Evolution (Grew & Hazen, 2009)



## Stage 9: Snowball Earth and Neoproterozoic Oxidation (0.85-0.542 Ga)

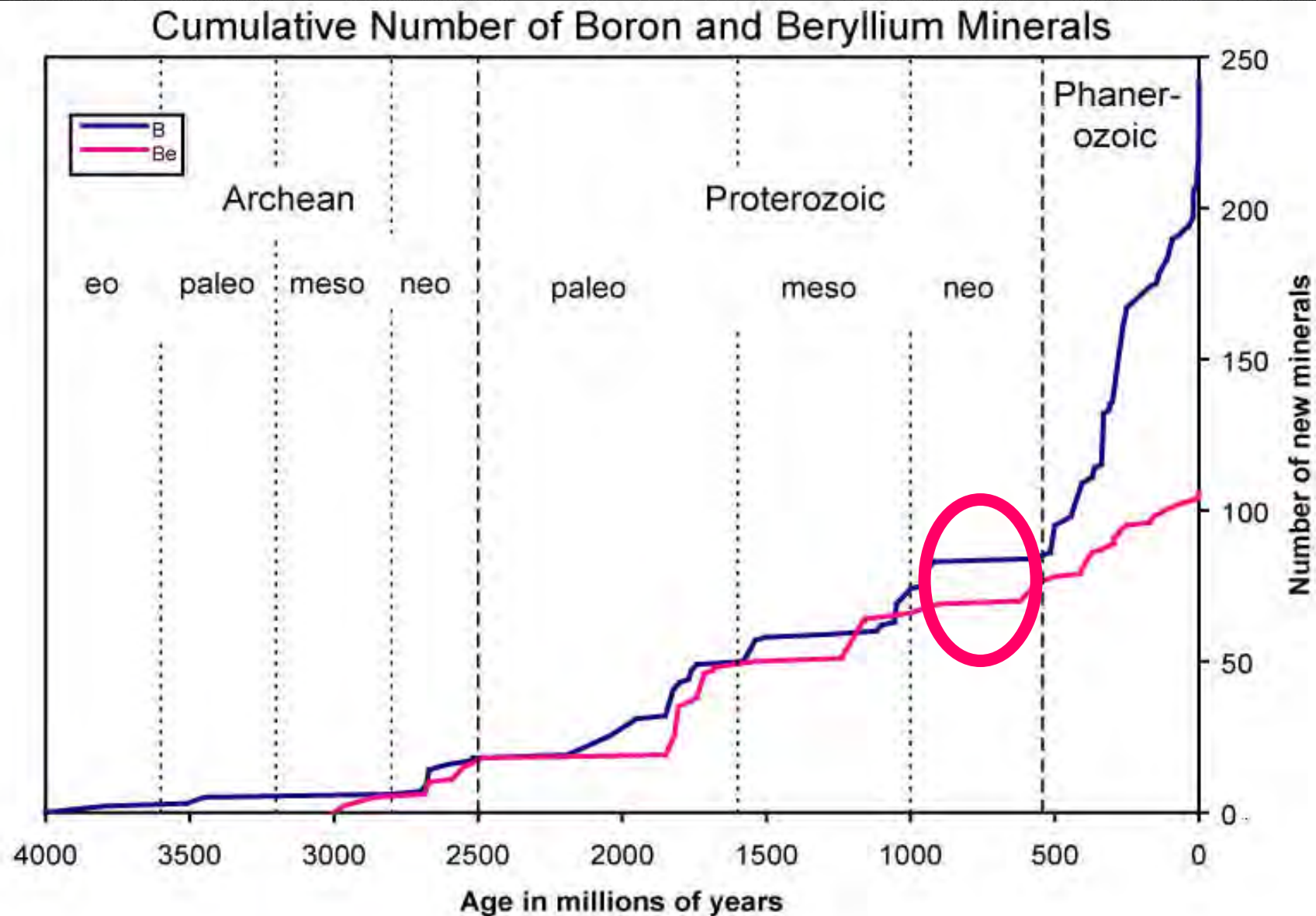
**>4000 mineral species (few new species)**



**Glacial cycles triggered by albedo feedback.**



# B Mineral Evolution (Grew & Hazen 2010)



## Stage 10: Phanerozoic Biomineralization ( $<0.542$ Ga)

**>4,400 mineral species (Biominerals, clays)**





# Stage 10: Phanerozoic Biomineralization ( $<0.542$ Ga)

**>4,400 mineral species**



# Stage 10: Phanerozoic Biomineralization



(c) Thomas Witzke + Abraxas Verlag

**Abelsonite— $\text{NiC}_{31}\text{H}_{32}\text{N}_4$**



**Ravatite— $\text{C}_{24}\text{H}_{48}$**

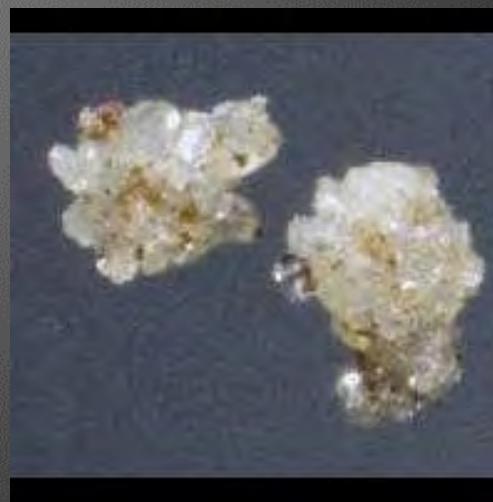


**Evankite— $\text{C}_{24}\text{H}_{48}$**



(c) Thomas Witzke + Abraxas Verlag

**Dashkovaite— $\text{Mg}(\text{HCOO})_2 \cdot 2\text{H}_2\text{O}$**

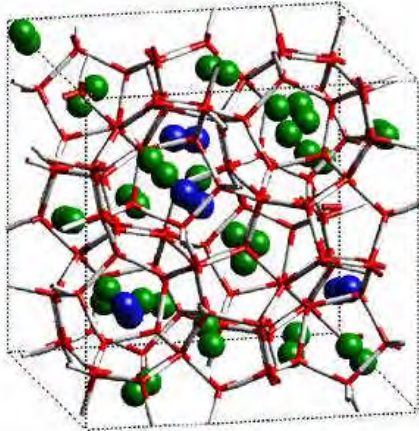


**Oxammite— $(\text{NH}_4)(\text{C}_2\text{O}_4) \cdot \text{H}_2\text{O}$**

**> 50 Organic Mineral Species**



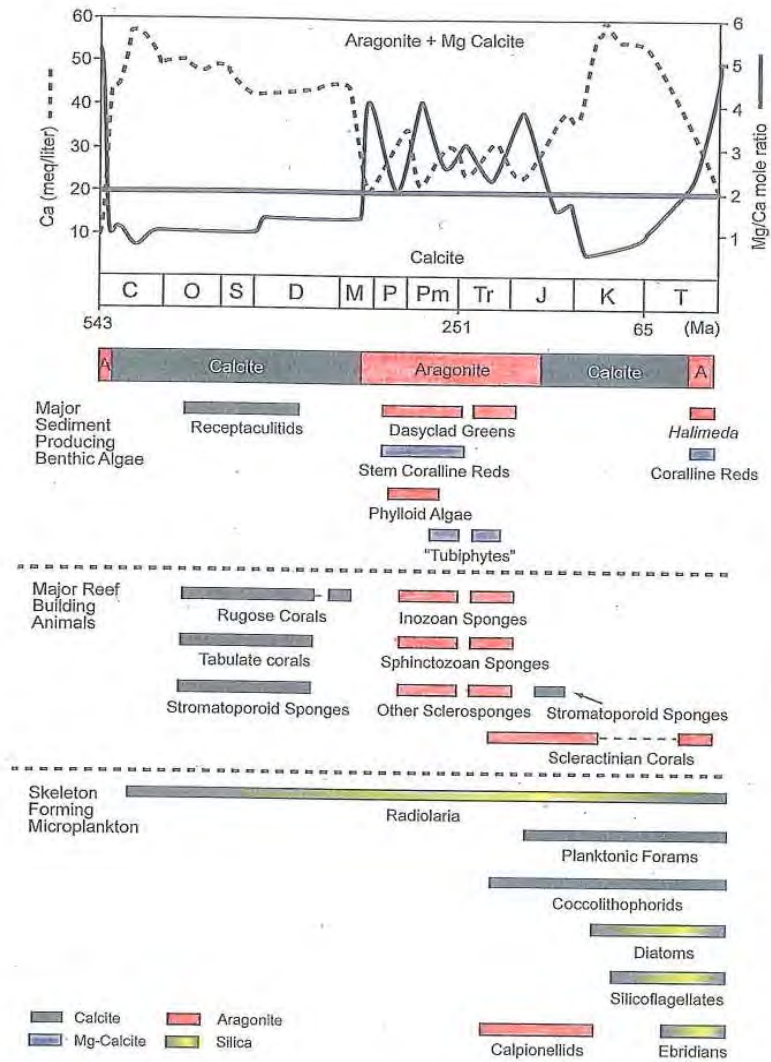
# Stage 10: Phanerozoic Biomineralization



Methane Ice— $\text{H}_2\text{O} \cdot n[\text{CH}_4]$

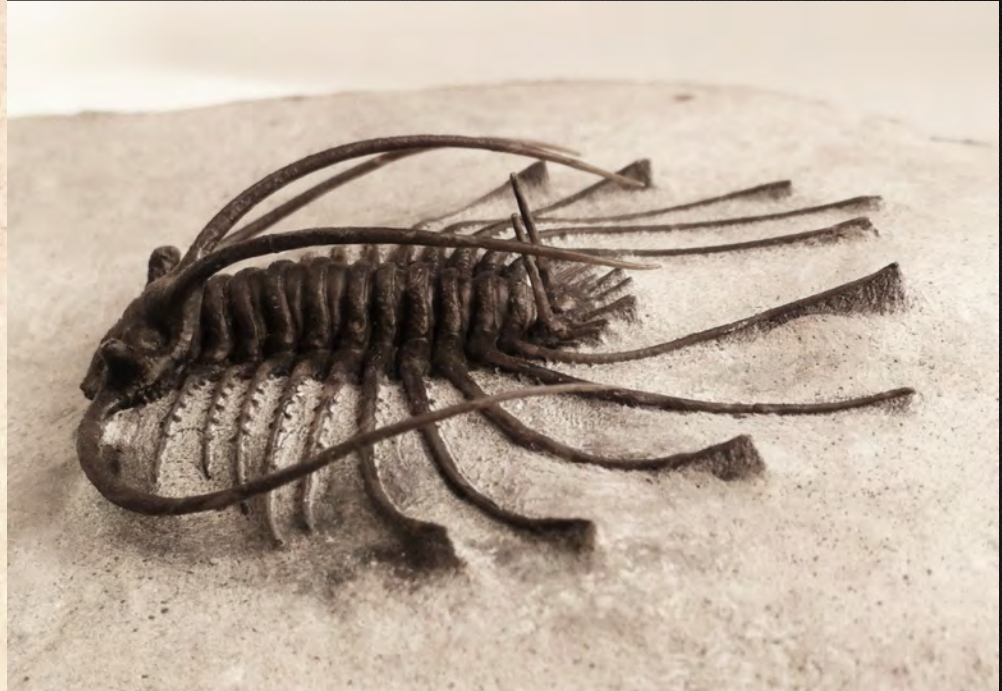


# Skeletal Biomineralization





## Stage 10: Phanerozoic Biomineralization

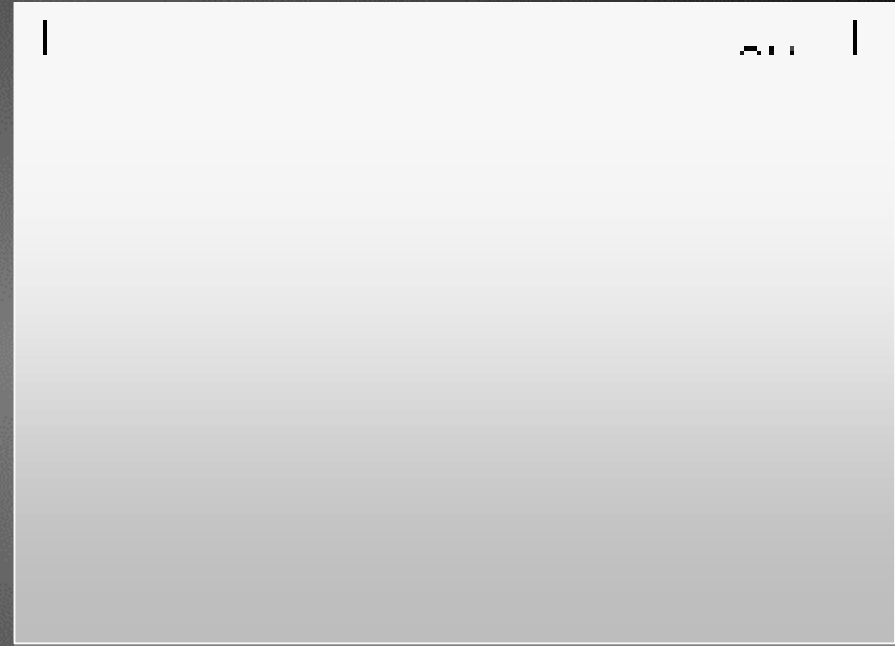
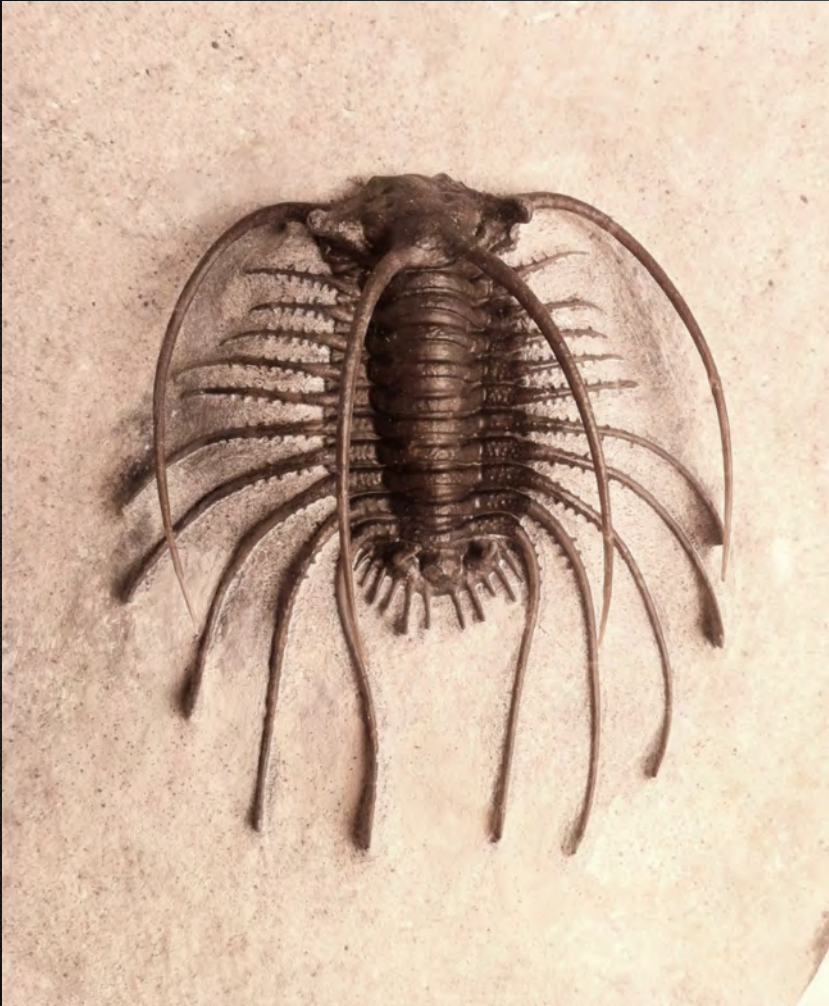


*Apianurus* nov. sp.

*Walcott-Rust Quarry,  
Moscow, New York.*



## Stage 10: Phanerozoic Biomineralization



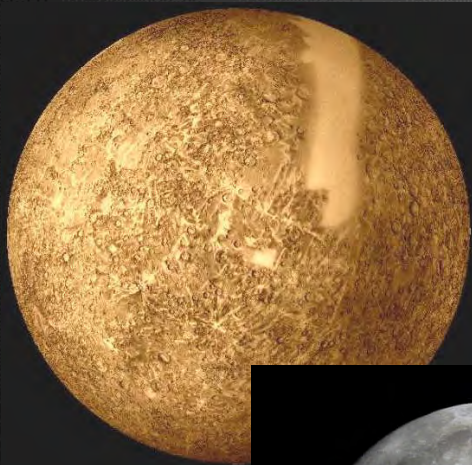
*Apianurus nov. sp.*

*Walcott-Rust specimens preserve biomolecular fragments of chitin.*



# Implications of Mineral Evolution

1. Mineral evolution suggests a new way to compare and contrast terrestrial planets and moons.



# **Implications of Mineral Evolution**

**2. Mineral evolution points to NASA mission targets: mineral biosignatures (and abiosignatures).**

- **Granites (pegmatites)**
- **Massive sulfide deposits**
- **Carbonates**
- **Banded iron formations**
- **Evaporites**



# Implications of Mineral Evolution

**3. Mineral evolution represents a new way to frame (and to teach) mineralogy.**

- **Provides a narrative thrust to the presentation of minerals.**

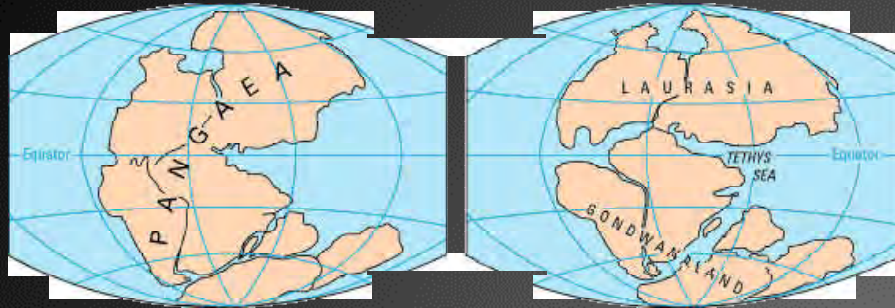
# RECENT CONCLUSIONS

Previously unrecognized patterns in the distribution of minerals through Earth history reflect:

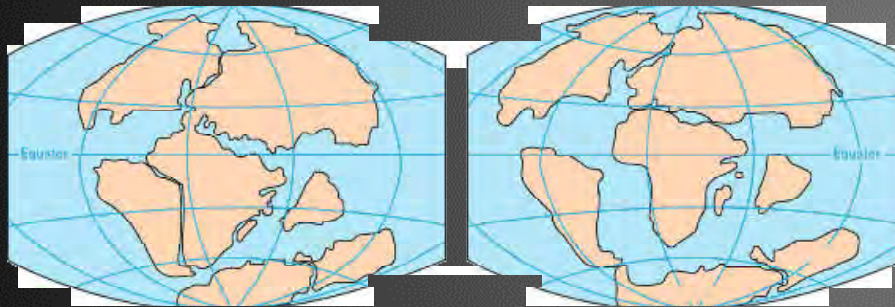
- The supercontinent cycle.
- Changes in Earth's near-surface oxidation state.
- Changing ocean chemistry.
- The rise of the terrestrial biosphere.



# The Supercontinent Cycle



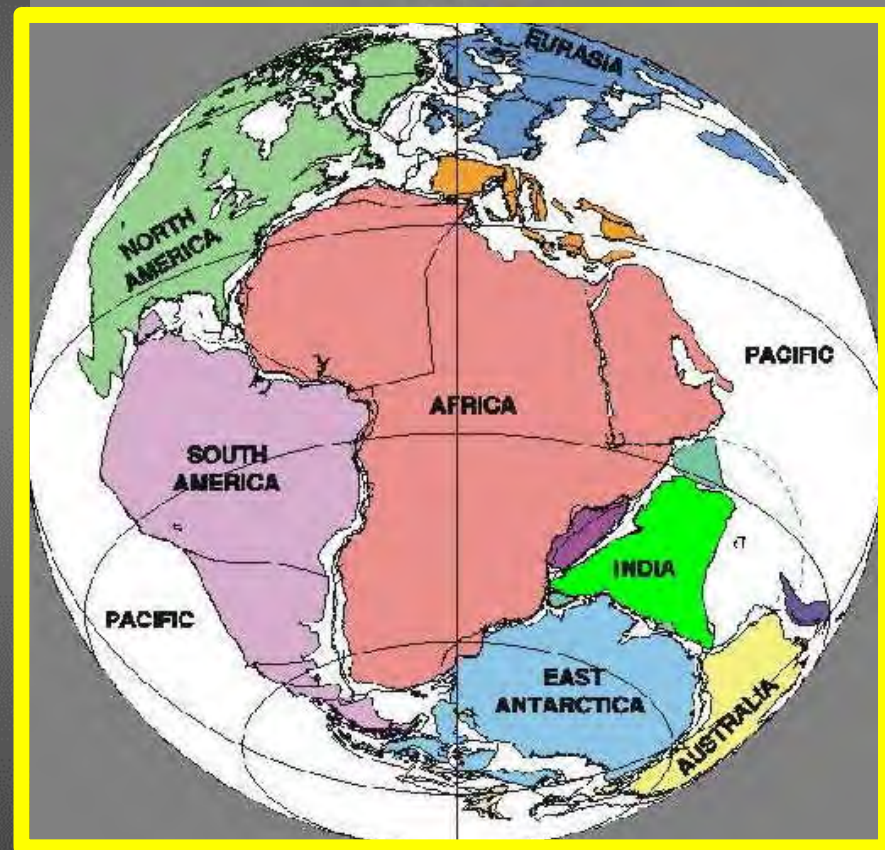
150 million years



150 million years



2011





# The Supercontinent Cycle

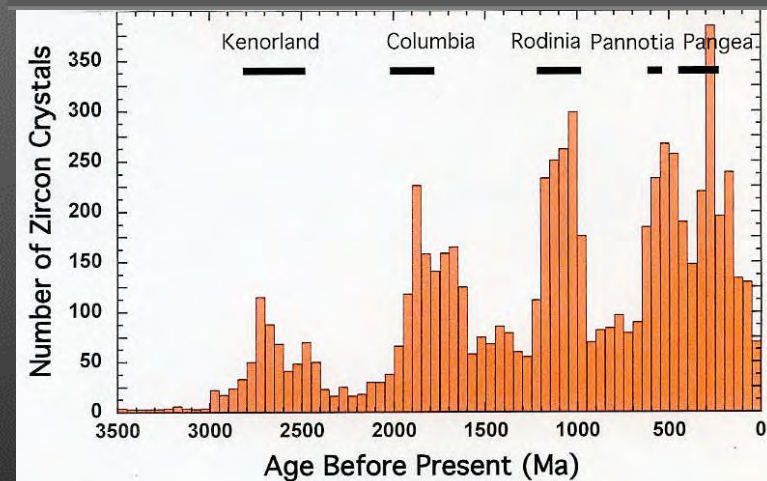
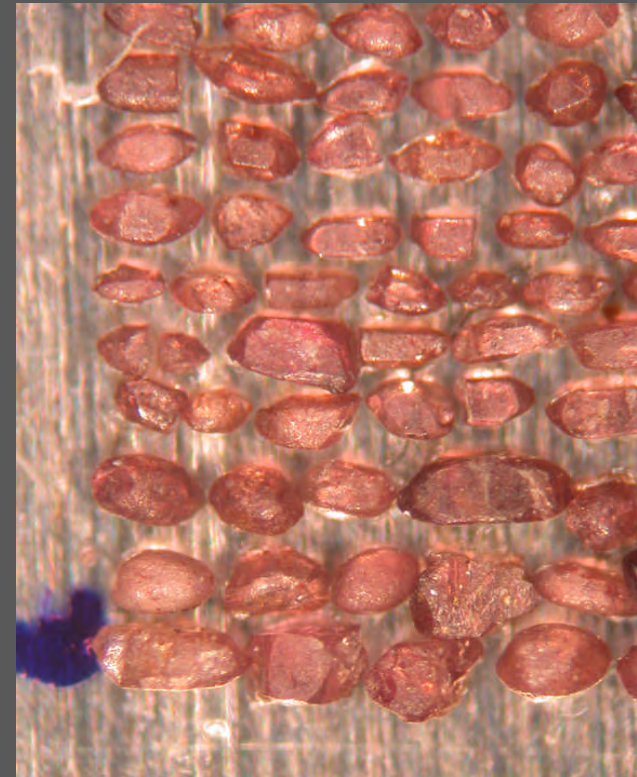
<b>SUPERCONTINENT</b>	<b>STAGE</b>	<b>INTERVAL</b>	<b>DURATION</b>
<b>Kenorland (Superia)</b>	<b>Assembly</b>	<b>2.8-2.5</b>	<b>300</b>
	<b>Stable</b>	<b>2.5-2.4</b>	<b>100</b>
	<b>Breakup</b>	<b>2.4-2.0</b>	<b>400</b>
<b>Columbia (Nuna)</b>	<b>Assembly</b>	<b>2.0-1.8</b>	<b>200</b>
	<b>Stable</b>	<b>1.8-1.6</b>	<b>200</b>
	<b>Breakup</b>	<b>1.6-1.2</b>	<b>400</b>
<b>Rodinia</b>	<b>Assembly</b>	<b>1.2-1.0</b>	<b>200</b>
	<b>Stable</b>	<b>1.0-0.75</b>	<b>250</b>
	<b>Breakup</b>	<b>0.75-0.6</b>	<b>150</b>
<b>Pannotia</b>	<b>Assembly</b>	<b>0.6-0.56</b>	<b>40</b>
	<b>Stable</b>	<b>0.56-0.54</b>	<b>20</b>
	<b>Breakup</b>	<b>0.54-0.43</b>	<b>110</b>
<b>Pangaea</b>	<b>Assembly</b>	<b>0.43-0.25</b>	<b>180</b>
	<b>Stable</b>	<b>0.25-0.175</b>	<b>75</b>
	<b>Breakup</b>	<b>0.175-present</b>	<b>175</b>



# RESULTS: The Supercontinent CYCLE

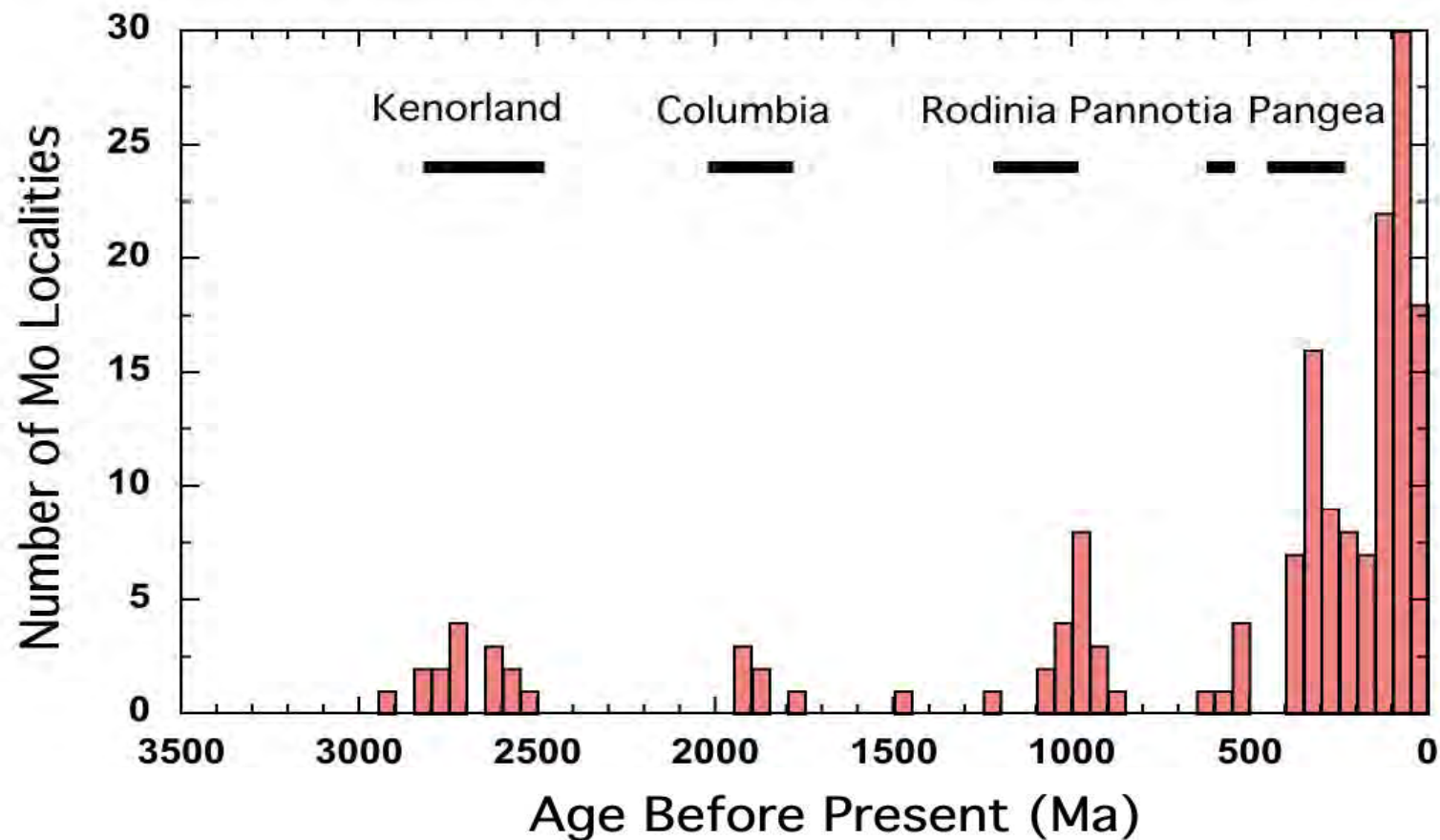
The distribution of zircon crystals through time correlates with the supercontinent cycle over the past 3 billion years.

(Condie & Aster 2010;  
Hawksworth et al. 2010)



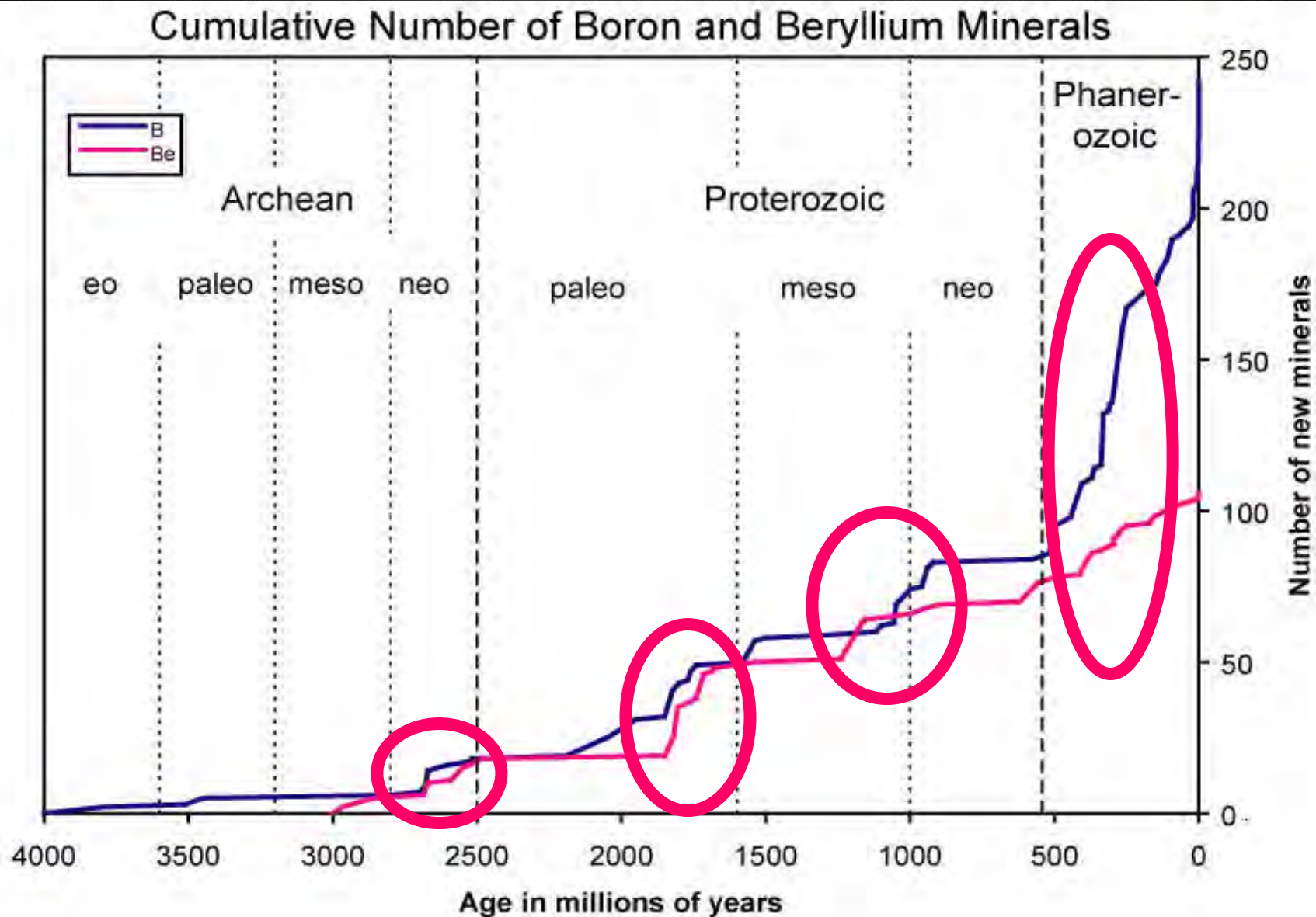
# RESULTS: Mo Mineral Evolution

## Temporal distribution of molybdenite ( $\text{MoS}_2$ )





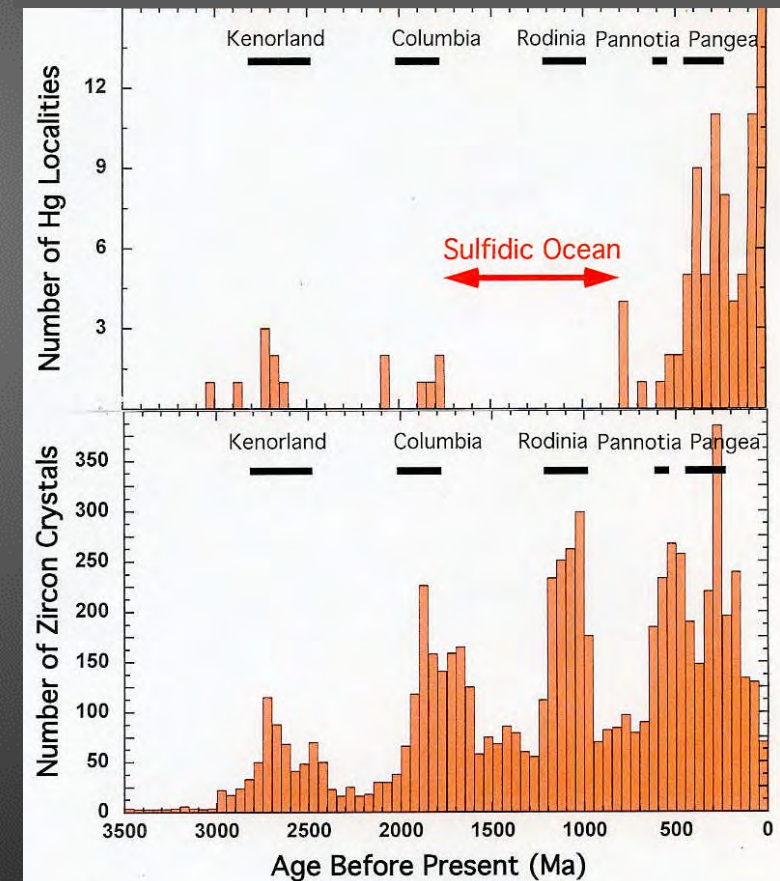
# B and Be Minerals (Grew & Hazen 2010)



# Hg Mineral Evolution

The distribution of mercury (Hg) minerals through time also correlates with the SC cycle over the past 3 billion years, but there's a gap during the “boring billion”.

Hazen et al. (2012) *Amer. Mineral.*, in press.

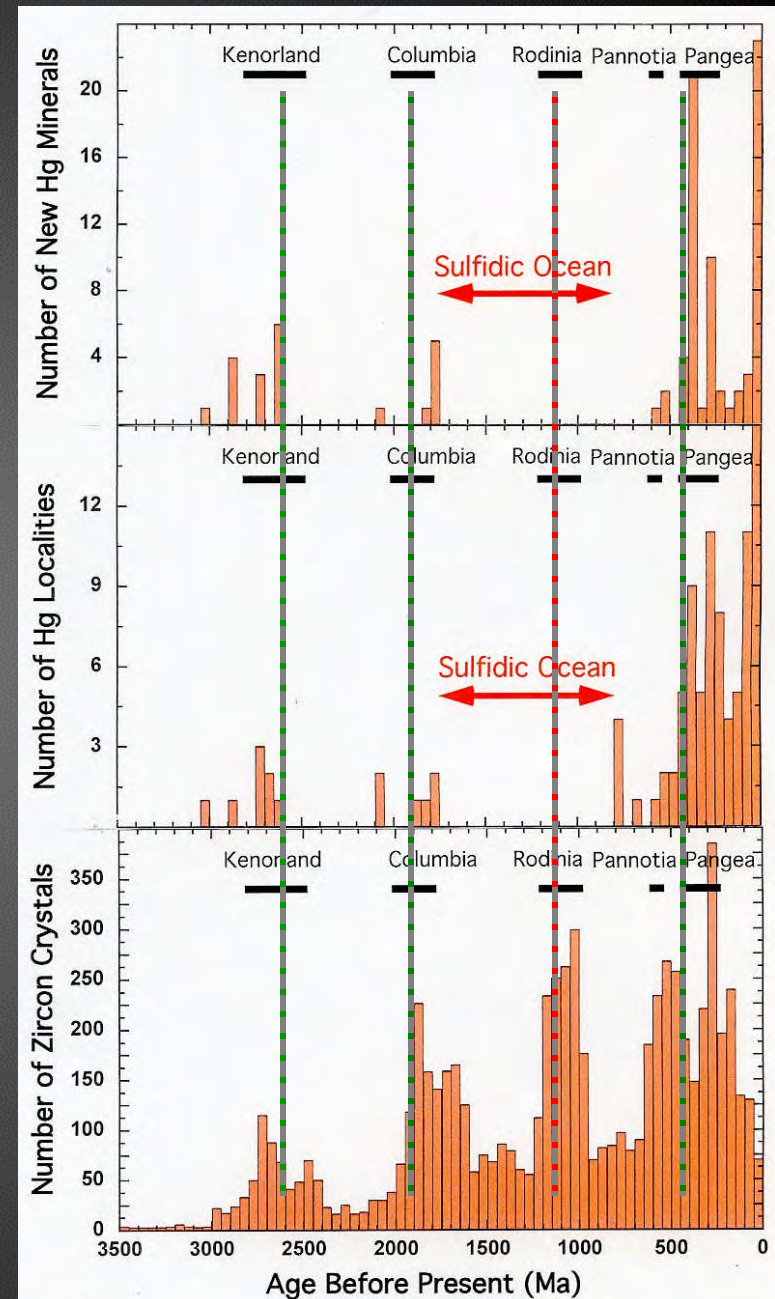




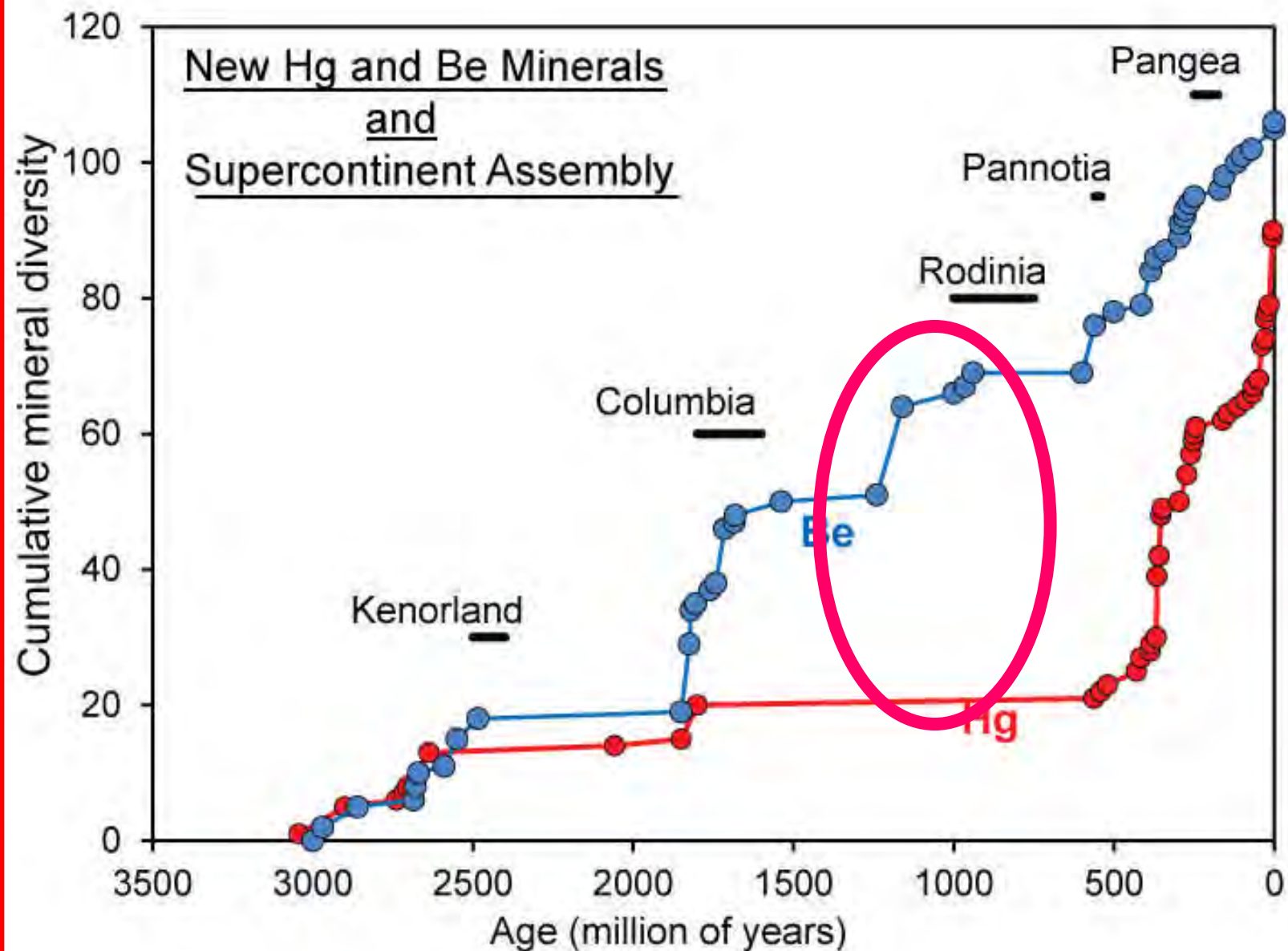
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# RESULTS: Hg & Be Mineral Evolution

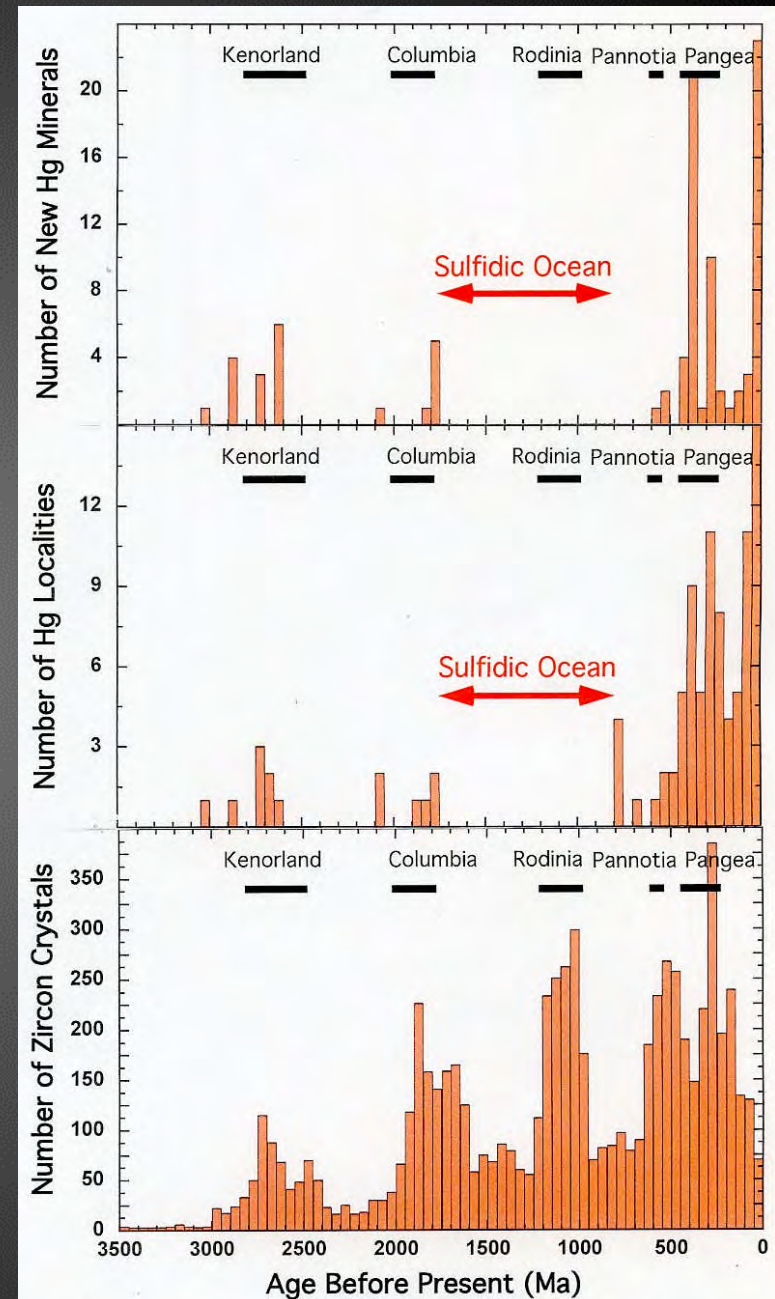




# Hg Mineral Evolution

We conclude that mercury was sequestered as insoluble cinnabar during the interval of the sulfidic ocean.

Hazen et al. (2012) *Amer. Mineral.*, in press.





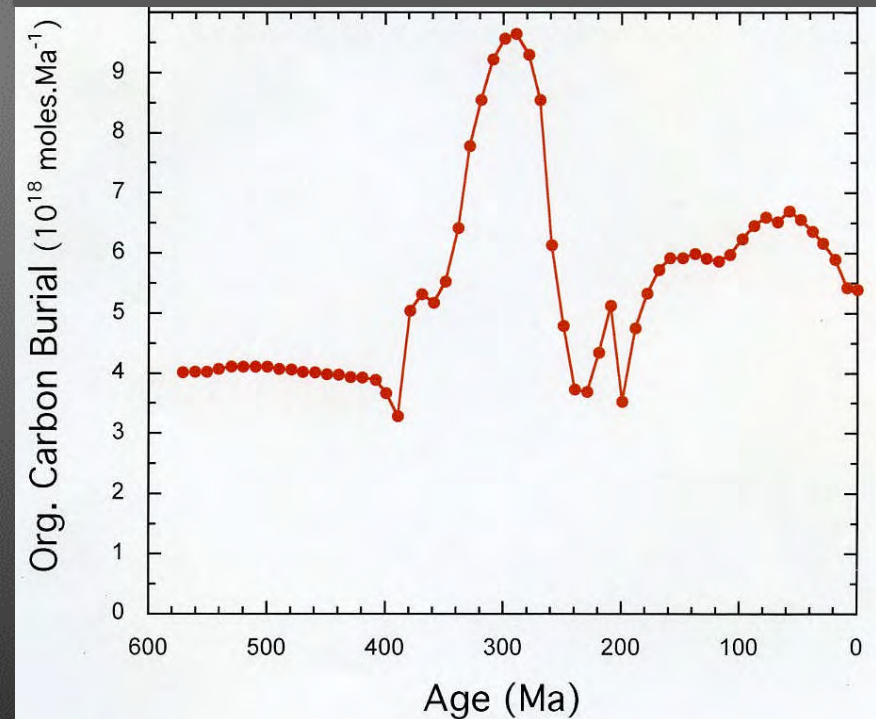
# Phanerozoic Biomineralization and the Terrestrial Biosphere (<542 Ma)





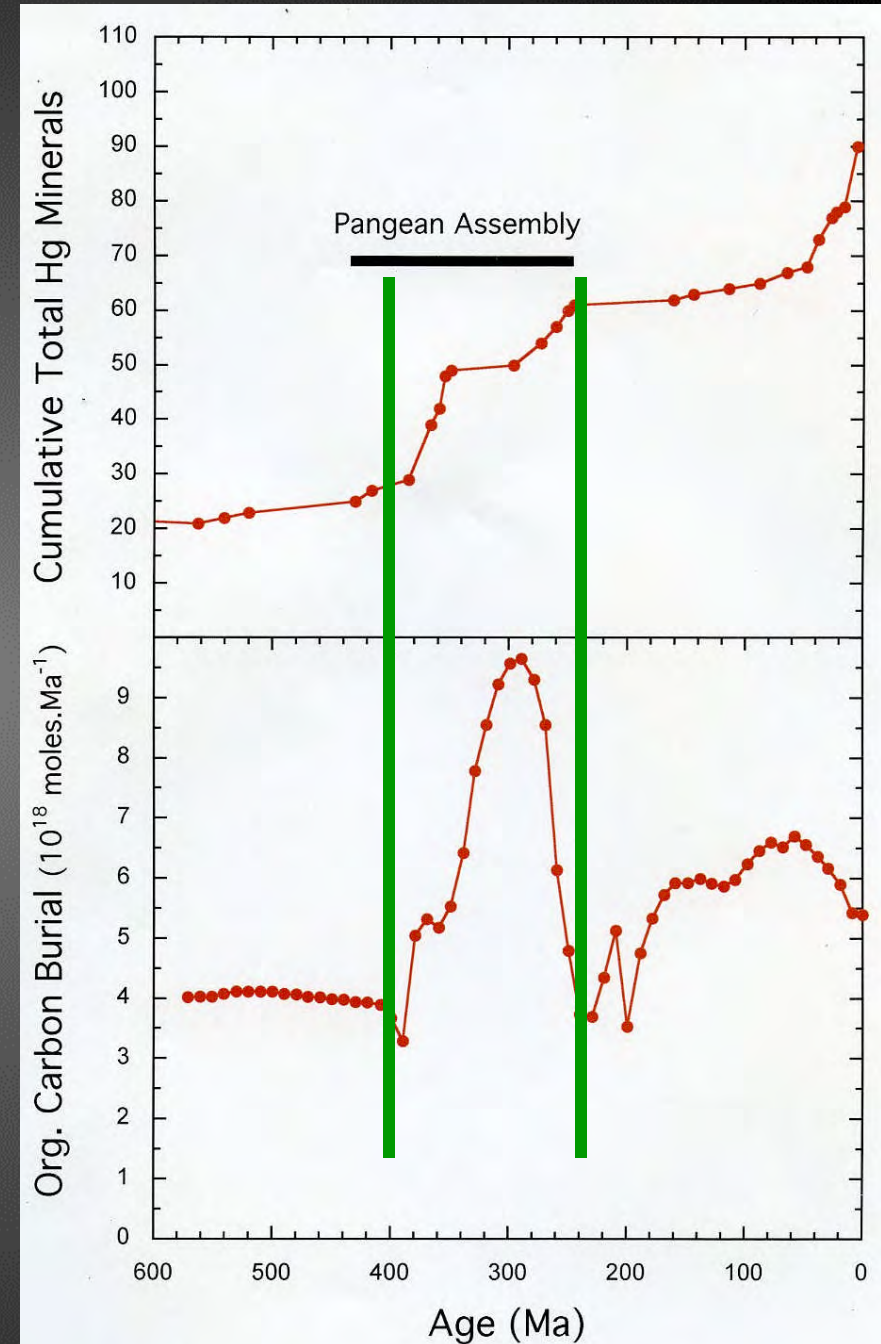
## Hg Mineral Evolution

The distribution of mercury (Hg) minerals through the last 400 million years reflects changes in Earth's biosphere.



# Hg Mineral Evolution

The distribution of mercury (Hg) minerals through the last 400 million years reflects changes in Earth's biosphere.





# Conclusions

- **The mineralogy of terrestrial planets and moons evolves in both deterministic and stochastic ways.**

Hazen et al. (2008) *Amer. Mineral.* 93, 1693; Hazen et al. (2009) *Amer. Mineral.* 94, 1293;  
Hazen et al. (2010) *Elements* 6, #1, 9-46; Hazen et al. (2011) *Amer. Mineral.* 96, 953.

# Conclusions

- The mineralogy of terrestrial planets and moons evolves in both deterministic and stochastic ways.
- **Different planets/moons achieve different stages of mineral evolution.**

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# Conclusions

- The mineralogy of terrestrial planets and moons evolves in both deterministic and stochastic ways.
- Different planets/moons achieve different stages of mineral evolution.
- **Three principal mechanisms of change:**
  1. **Element segregation & concentration**
  2. **Increasing ranges of T, P and X**
  3. **Influence of living systems.**

Hazen et al. (2008) *Amer. Mineral.* 93, 1693; Hazen et al. (2009) *Amer. Mineral.* 94, 1293;  
Hazen et al. (2010) *Elements* 6, #1, 9-46; Hazen et al. (2011) *Amer. Mineral.* 96, 953.

# Conclusions

**With mineral evolution, the science of mineralogy once again assumes its rightful place at the center of the Earth and planetary sciences.**





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